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BENZOTHIAZINE DERIVATIVE

Φ A benzothiazine derivative represented by general formula (I) or a salt thereof, wherein the broken line indicates the presence or absence of a bond; Z represents any of the groups (a), (b), (c), (d), (e), (f), (g) and (h); (wherein R₁ represents alkyl, aralkyl, etc.; R₂ represents alkyl, aralkyl, etc.; R₃ represents H, alkyl, etc.; R₃ represents H, alkyl, etc.; R₃ represents H, aralkyl, etc.; X₁, X₂ and X₃ represent each O or S; and G represents optionally substituted ethylene, trimethylene, etc.); A represents alkylene, alkenylene, etc.; Y represents CH, C = or N, provided when Y is CH, then m represents 0 or 1, n represents 1 or 2, and B represents 0, S, carbonyl, etc., when Y is C =, then m represents 1, n represents 1 or 2, and B represents (i) (wherein the double bond is bound to Y, and R₅ represents optionally substituted aryl, etc.), and when Y is N, then m represents 0 or 1, n represents 2 or 3, and B represents carbonyl, sulbonyl, etc.; E₁ and E₂ represent each H or lower alkyl; and D represents an aromatic hydrocarbon group, aromatic heterocyclic group, etc. The benzothiazine derivatives and salts thereof have a potent serotonin 2 blocker activity and an excellent α₁ blocker activity selectivity and are highly safe, thus providing drugs utilizing a serotonin 2 receptor antagonism, for example, remedies for circulatory diseases such as ischemic heart diseases, cerebrovascular diseases and peripheral circulation disorder.

Technical Field

This invention relates to novel benzothiazine derivatives. More specifically, this invention is concerned with benzothiazine derivatives and salts thereof, said derivatives and salts being useful for the prevention or treatment of ischemic heart diseases such as angina pectoris, arrhythmia, myocardial infarction, congestive heart failure and post-PTCA restenosis, cerebrovascular disturbances such as cerebral infarction and cerebral sequelae after subarachnoid hemorrhage, and/or peripheral circulatory disturbances such as arteriosclerosis obliterans, Raynaud disease, Buerger disease and thrombophlebitis; their preparation process; and therapeutics for circulatory diseases, said therapeutics containing them as effective ingredients.

Background Art

Serotonin is a compound contained abundantly in platelets, which are a blood component, and in a central nervous system, it acts as a neurotransmitter. In platelets, it is released upon stimulation by thromboxane A₂, ADP, collagen or the like and synergistically acts on various platelet aggregation factors and vasoconstrictors through activation of serotonin-2 receptors in the platelets and vascular smooth muscle cells, thereby inducing strong platelet aggregation and vasoconstriction [P.M. Vanhoutte, "Journal of Cardiovascular Pharmacology", Vol. 17 (Supple. 5), S6-S12 (1991)].

Serotonin is also known to potentiate proliferation of vascular smooth muscle cells [S. Araki et al., "Atherosclerosis", Vol. 83, p29-p34(1990)]. It has been considered that, particularly when endothelial cells are injured as in arteriosclerosis or myocardial infarction, the vasoconstricting action and thrombus forming action of serotonin are exasperated, thereby reducing or even stopping blood supply to myocardial, cerebral and peripheral organs [P. Golino et al., "The New England Journal of Medicine", Vol. 324, No. 10, p641-p648(1991), Y. Takiguchi et al., "Thrombosis and Haemostasis", Vol. 68(4), p460-p463(1992), A.S. Weyrich et al., "American Journal of Physiology", Vol. 263, H349-H358(1992)].

Being attracted by such actions of serotonin or serotonin-2 receptors, various attempts are now under way to use a serotonin-2 receptor antagonist as a pharmaceutical for ischemic diseases of the heart, the brain and peripheral tissues.

Ketanserin which has therapeutically been used as a hypotensive drug is known as a compound having antagonistic action against a serotonin-2 receptor. Ketanserin has strong antagonistic action against a sympathetic nerve α_1 receptor and also against histamine 1 and dopamine receptors in addition to as antagonistic action against serotonin-2 receptors, so that there is the potential problem of developing excessive hypotensive action, neuroleptic action or the like when used for the treatment of ischemic heart disease or peripheral circulatory disturbance. Ketanserin is therefore not preferred.

In addition, several compounds, including sarpogrelate, are known to have serotonin-2 receptor antagonistic action. They are however accompanied with problems in the potency, the selectivity to other receptors, toxicity, side effects or the like. Thus, there remains still much room for improvements.

Disclosure of the Invention

In view of the foregoing circumstances, the present inventors synthesized various compounds and investigated their pharmacological effects. As a result, it has been found that specific benzothiazine derivatives have strong serotonin-2 receptor antagonistic action, are excellent in the selectivity to a serotonin-2 receptor in the antagonistic action against various receptors, particularly in the selectivity to a serotonin-2 receptor in the antagonistic action against α_1 receptors, and have low toxicity, leading to the completion of the present invention.

The present invention has been completed based on the above described findings. A first object of the present invention is to provide a benzothiazine derivative represented by the following formula (I):

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wherein the dashed line indicates the presence or absence of a bond, and when the dashed line indicates the presence of the bond, Z represents one of the following groups:



in which R₁ represents a substituted or unsubstituted alkyl group or a substituted or unsubstituted aralkyl group but, when the dashed line indicates the absence of the bond, Z represents one of the following groups:

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in which R₂ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, R₃ represents a hydrogen atom, a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, R₄ represents a hydrogen atom, a substituted or unsubstituted alkyl group or a substituted or unsubstituted aralkyl group, X₁, X₂ and X₃ each independently represents an oxygen atom or a sulfur atom, G represents an ethylene group with one or more of the hydrogen atoms thereof optionally substituted by a like number of halogen atoms and/or alkyl, aryl, aralkyl and/or alkylidene groups, or a trimethylene group with one or more of the hydrogen atoms thereof optionally substituted by a like number of halogen atoms and/or alkyl, aryl, aralkyl and/or alkylidene groups,

A represents a substituted or unsubstituted alkylene group, a substituted or unsubstituted alkenylene group or a substituted or unsubstituted alkynylene group,

Y represents CH, C = or a nitrogen atom and when Y represents CH, m stands for 0 or 1, n stands for 1 or 2, B represents an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an alkylene group, an alkenylene group, a substituted or unsubstituted hydroxymethylene group, a group -CHR₅- in which R₅ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, or a substituted or unsubstituted cyclic or acyclic acetal group, when Y represents C = , m stands for 1, n stands for 1 or 2, B represents a group:

$$=$$
 R_6

in which the double bond is linked to Y, R₆ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, but when Y represents a nitrogen atom, m stands for 0 or 1, n stands for 2 or 3, B represents a carbonyl group, a sulfonyl group, an alkylene group, an alkenylene group, or a group -CHR₇-in which R₇ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group,

E₁ and E₂ each independently represents a hydrogen atom or a lower alkyl group and,

D represents a substituted or unsubstituted aromatic hydrocarbon group or a substituted or unsubstituted aromatic heterocyclic group; or a salt thereof.

Another object of the present invention is to provide a preparation process of the benzothiazine derivative (I) or its salt.

A further object of the present invention is to provide a serotonin-2 receptor antagonist as a treating agent for circulatory diseases or the like, said antagonist containing the benzothiazine derivative (I) or its pharmaceutically-acceptable salt as an effective ingredient.

Best Modes for Carrying Out the Invention

In the benzothiazine derivatives (I) of the present invention, preferred examples of group R_1 include branched or linear C_{1-4} alkyl groups such as methyl and ethyl and C_{7-22} aralkyl groups such as benzyl and phenethyl, each of which may be substituted by one or more of halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl; and/or alkoxy groups, preferably C_{1-4} alkoxy groups such as methoxy and ethoxy.

Preferred examples of group R_2 include branched or linear C_{1-4} alkyl groups such as methyl and ethyl, C_{6-14} aryl groups such as phenyl and naphthyl and C_{7-22} aralkyl groups such as benzyl and phenethyl, each of which may be substituted by one or more of halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl; and/or alkoxy groups, preferably C_{1-4} alkoxy groups such as methoxy. In this case, preferred examples of group R_2X_1 - include methoxy, methylthio, ethoxy and ethylthio groups.

Preferred examples of the group:

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 x_2 x_3

includes groups represented by the following formulas:

0 0 0 0 S S S S O S O S

C , C , C , C , C and C

in which one or more of the hydrogen atoms may be substituted by a corresponding number of halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl; aryl groups, preferably C_{6-14} aryl groups such as phenyl and naphthyl; aralkyl groups, preferably C_{7-22} aralkyl groups such as benzyl and phenethyl; and/or alkylidene groups, preferably C_{1-4} alkylidene groups such as methylidene and ethylidene.

Preferred examples of group R_3 of group NOR $_3$ include a hydrogen atom, branched or linear C_{1-4} alkyl groups such as methyl and ethyl C_{6-14} aryl groups such as phenyl and naphthyl and C_{7-22} aralkyl groups such as benzyl and phenethyl. Each of the exemplified groups may be substituted by one or more of halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl; and/or alkoxy groups, preferably C_{1-4} alkoxy groups such as methoxy and ethoxy.

Preferred examples of group R_4 include a hydrogen atom, branched or linear C_{1-4} alkyl groups such as methyl and ethyl and C_{7-22} aralkyl groups such as benzyl and phenethyl. Each of the exemplified groups may be substituted by one or more of halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl; and/or alkoxy groups, preferably C_{1-4} alkoxy groups such as methoxy and ethoxy.

Preferred examples of group Z include groups:

Specifically preferred examples of group Z include groups:

 R_2X_1 X_1R_2 X_2 X_3 OR_1 H OR_2 C , C and C

wherein G, R₁, R₂, R₄, X₁, X₂ and X₃ have the same meanings as defined above.

Preferred examples of group A include branched or linear C_{2-10} alkylene groups such as ethylene, trimethylene, tetramethylene, pentamethylene and octamethylene, branched or linear C_{4-10} alkenylene groups such as 2-butenylene and 3-pentenylene, and branched or linear C_{4-10} alkynylene groups such as 2-butynylene and 3-pentynylene. Each of the exemplified group may be substituted by one or more of halogen atoms such as fluorine, chlorine and bromine. Among them, ethylene, trimethylene and tetramethylene groups are particularly preferred.

The following group:

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$$-\underbrace{\overset{(CH_2)_n}{\underset{E_1}{\bigvee}}_{Y}}_{E_1}$$

wherein E_1 , E_2 , Y and n have the same meanings as defined above, is a heterocyclic group led by a pyrrolidine, piperazine or homopiperazine group, in which two or less hydrogen atoms on the ring may be substituted by lower alkyl groups, preferably C_{1-4} alkyl groups such as methyl or ethyl.

When the group of the above formula is a heterocyclic group led by a pyrrolidine or piperidine group, preferably a piperidine group, m stands for 0 or 1 (m stands for 1 with the proviso that Y represents C =), B represents an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an alkylene group (preferably a C_{1-4} alkylene group and most preferably a 2-propenylene group), an alkenylene group (preferably C_{2-5} alkenylene group and most preferably a 2-propenylene group), a substituted or unsubstituted hydroxymethylene group, a group -CHR_S- (in which R_S represents an alkyl group, preferably a C_{1-4} alkyl group such as methyl and ethyl; an aryl group, preferably C_{6-14} aryl group such as phenyl and naphthyl; or an aralkyl group, preferably C_{7-22} aralkyl group such as benzyl or phenethyl), a group represented by the following formula:

wherein the double bond is linked to Y, R_6 represents an alkyl group, preferably a C_{1-4} alkyl group such as methyl and ethyl; an aryl group, preferably C_{6-14} aryl group such as phenyl and naphthyl; and an aralkyl group, preferably a C_{7-22} aralkyl group such as benzyl and phenethyl, which may be substituted or unsubstituted, or a cyclic acetal or acyclic acetal group in which one or more of hydrogen atoms may be substituted.

Exemplary cyclic or acyclic acetal groups include groups represented by the following formulas:

Among those exemplified above as B, preferred examples of the substituent for a hydroxymethylene group include alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl, and aryl groups, preferably C_{6-14} aryl groups such as phenyl and naphthyl, each group being substituted for a carbon atom of the methylene group. These substituents may each be substituted further by one or more substituents such as hydroxyl groups, halogen atoms, e.g., fluorine, chlorine and/or bromine and/or alkoxyl groups, preferably C_{1-4} alkoxyl groups such as methoxy and ethoxy.

Particularly preferred examples of the substituted or unsubstituted hydroxymethylene group include an unsubstituted hydroxymethylene group and a hydroxymethylene group substituted by a phenyl, fluorophenyl or hydroxyphenyl group.

Examples of the substituent for R_5 include one or more of halogen atoms such as fluorine, chlorine and bromine, alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl, and alkoxyl groups, preferably C_{1-4} alkoxyl groups such as methoxy and ethoxy.

Exemplary substituents for R_6 include one or more of halogen atoms such as fluorine, chlorine and bromine, alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl, alkoxyl groups, preferably C_{1-4} alkoxyl groups such as methoxy and hydroxyl groups. Illustrative of the substituent for the cyclic or acyclic acetal include halogen atoms such as fluorine, chlorine and bromine, alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl, aryl groups, preferably C_{6-14} aryl groups such as phenyl and naphthyl, aralkyl groups, preferably C_{7-22} aralkyl groups such as benzyl and phenethyl and alkylidene groups, preferably C_{1-4} alkylidene groups such as methylidene and ethylidene.

Among those exemplified above as B, a carbonyl group is most preferred.

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When the heterocyclic group is a group led by a piperazine or homopiperazine group, preferably a piperazine group, m stands for 0 or 1 (preferably 0), B represents a carbonyl group, a sulfonyl group, an alkylene group (preferably a C_{1-4} alkylene group, particularly a methylene group), an alkenylene group (preferably C_{3-6} alkenylene group, particularly 2-propenylene group) or a group -CHR₇- (in which R₇ represents an alkyl group, preferably a C_{1-4} alkyl group such as methyl and ethyl; an aryl group, preferably C_{6-14} aryl group such as phenyl and naphthyl; or an aralkyl group, preferably a C_{7-22} aralkyl group such as benzyl and phenethyl).

Further, R_7 may be substituted, for example, by one or more of halogen atoms such as fluorine, chlorine and bromine, alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl, and/or alkoxy groups, preferably C_{1-4} alkoxy groups such as methoxy and ethoxy.

Among those exemplified above as B, preferred are substituted or unsubstituted phenylmethylene groups.

Preferred examples of group D include aromatic hydrocarbon groups, preferably C_{6-28} aromatic hydrocarbon groups such as a phenyl group in which one or more of the hydrogen atoms may be substituted and a naphthyl group in which one or more of the hydrogen atoms may be substituted. Other preferred examples of D include aromatic heterocyclic groups, preferably those each of which is monocyclic or dicyclic and contains three or less oxygen, sulfur and/or nitrogen atoms - such as pyridyl, pyrimidyl, benzisothiazolyl, benzisoxazolyl, indazolyl and indolyl groups in which one or more of hydrogen atoms may be substituted.

Examples of the substituents for the above aromatic hydrocarbon or aromatic heterocyclic group include halogen atoms such as fluorine, chlorine and bromine; alkyl groups, preferably C_{1-4} alkyl groups such as methyl and ethyl; alkoxyl groups, preferably C_{1-4} alkoxyl groups such as methoxy and ethoxy; aryl groups, preferably C_{6-14} aryl groups such as phenyl and naphthyl; aralkyl groups, preferably C_{7-22} aralkyl groups, such as benzyl and phenethyl; aralkyloxy groups, preferably C_{7-22} aralkyloxy groups such as benzyloxy; cyano groups; nitro groups; carboxyl groups; alkoxycarbonyl groups (having preferably a C_{1-6} alcohol moiety); lower alkylsulfonylamino groups (having preferably a C_{1-4} alkyl moiety); carbamoyl groups; and hydroxyl groups.

Preferred examples of group D among these examples include phenyl groups which have been unsubstituted or substituted by one or more of halogen atoms, alkoxy groups and/or hydroxyl groups; benzisothiazolyl groups which have been unsubstituted or substituted by one or more halogen atoms; benzisoxazolyl groups which have been unsubstituted or substituted by one or more halogen atoms; and indazolyl groups which have been unsubstituted or substituted by one or more halogen atoms. Particularly

preferred are phenyl groups which have been unsubstituted or substituted by one or more of fluorine atoms, methoxy groups and/or hydroxyl groups.

Many of the compounds (I) according to the present invention have isomers. It is to be noted that these isomers and mixtures thereof are all embraced by the present invention.

Various processes can be employed for the preparation of the benzothiazine derivatives (I) according to the present invention. It is however preferred to prepare the benzothiazine derivatives, for example, by any one of the following processes.

Process 1:

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Among the benzothiazine derivatives (I), each of the compounds (Ib) in which Z is represented by a group:

$$R_2X_1$$
 X_1R_2 X_2 X_2 X_3

can be synthesized in accordance with any of the processes shown by the following schemes.

(a) The intended compound can be obtained, in accordance with the following reaction scheme, by reacting the compound represented by formula (XXII) with the compound represented by formula (III) to convert the former compound into the compound represented by formula (XXIII) and then reacting the compound (XXIII) with a nitrogen-containing compound represented by formula (V) or a salt thereof.

wherein A, B, D, E_1 , E_2 , Y, m and n have the same meanings as defined above; Z_2 represents one of the following groups:

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$$R_2X_1$$
 X_1R_2 X_2 X_3

in which G, R_2 , X_1 , X_2 and X_3 have the same meanings as defined above; and W and W' may be the same or different and individually represents a substituent easily replaceable with an amino group.

In the above reaction, the conversion from the compound (XXII) to the compound (XXIII) can be effected by treating the compound (XXII) with an organic or inorganic base and then causing the compound (III) to act on the compound (XXII), or by causing the compound (III) to act on the compound (XXII) in the presence of such a base.

Examples of group W or W' of the compound (III), which is an eliminative substituent and easily replaceable with an amino group, include halogen atoms such as chlorine and bromine, alkylsulfonyloxy groups such as methanesulfonyloxy and arylsulfonyloxy groups such as p-toluenesulfonyloxy.

Exemplary organic or inorganic bases include sodium hydride, potassium hydride, sodium carbonate, potassium carbonate, triethylamine and potassium t-butoxide. Further, illustrative solvents useful for the above reaction include tetrahydrofuran, dioxane, dimethylformamide, dimethylsulfoxide, acetonitrile, N-methylpyrrolidone, acetone, 2-butanone and toluene. The reaction is conducted at -78 °C to reflux temperature.

To prepare the compound (Ib) by reacting the compound (XXIII) with the nitrogen-containing compound (V), it is only necessary to react the nitrogen-containing compound (V) or an organic acid or inorganic acid salt thereof with the compound (XXIII), optionally together with an organic base such as triethylamine, pyridine, collidine, 1,8-diazabicyclo[5.4.0]undec-7-en (DBU) or potassium t-butoxide or an inorganic base such as potassium carbonate, sodium carbonate, sodium hydrogencarbonate, sodium hydroxide or sodium hydride, optionally after adding an alkali iodide such as potassium iodide or sodium iodide, at 0 °C to 150 °C in the solvent exemplified above or a solvent such as methanol, ethanol, propanol or butanol.

Examples of the nitrogen-containing compound (V) include 1-phenylpiperazine, 1-(2-fluorophenyl)piperazine, 1-(3-fluorophenyl)piperazine, 1-(4-fluorophenyl)piperazine, 1-(4-hydroxyphenyl)piperazine, 1-(2-chlorophenyl)piperazine, 1-(3-chlorophenyl)piperazine, 1-(4-chlorophenyl)piperazine, 1-(2-methoxyphenyl)piperazine, 1-(3-methoxyphenyl)piperazine, 1-(4-methoxyphenyl)piperazine, 1-(4-methanesulfonamidophenyl)piperazine, 1-(4-cyanophenyl)piperazine, 1-(4-carbamoylphenyl)piperazine, 1-(4-methoxycarbonylphenyl)piperazine, 1-(2-pyridyl)piperazine, 1-(2-pyrimidyl)piperazine, 1-benzylpiperazine, 1diphenylmethylpiperazine, 1-cinnamylpiperazine, 1-benzoylpiperazine, 1-(4-benzyloxybenzoyl)piperazine, 1-(4-hydroxybenzoyl)piperazine, 1-(2-furoyl)piperazine, 1-(1,2-benzisoxazol-3-yl)piperazine, 1-(1,2-ben-4-phenylpiperidine, 4-benzylpiperidine, α,α -bis(4-fluorophenyl)-4zisothiazol-3-yl)piperazine, piperidinemethanol, 4-(4-fluorobenzoyl)piperidine, 4-benzoylpiperidine, 4-(4-methoxybenzoyl)piperidine, 4-(4-chlorobenzoyl)piperidine, 3-(4-fluorobenzoyl)piperidine, 4-(6-fluoro-1,2-benzisoxazol-3-yl)piperidine, 4-(6-fluoro-1,2-benzisothiazol-3-yl)piperidine, 4-(6-fluoro-1H-indazol-3-yl)piperidine, 3-benzoylpyrrolidine, 3-(4-fluorobenzoyl)pyrrolidine, 4-(4-fluorophenoxy)piperidine, 4-[(4-fluorophenyl)thio]piperidine, 4-[(4fluorophenyl)sulfinyl]piperidine, 4-[(4-fluorophenyl)sulfonyl]piperidine, 4-[bis(4-fluorophenyl)methylene]piperidine and 4-(4-fluorobenzoyl)piperidine ethylene acetal. They are all either known compounds or compounds which can be readily prepared by a known process or a process similar to the known process.

Incidentally, among the compounds (XXII) employed as starting materials in the above reaction, the compounds except those in which Z₂ is a group represented by the following formula:

are novel compounds. These novel compounds can each be prepared from a known compound (XV) by any one of the various processes. For example, it is possible to select a suitable method among those

described in "Protective Groups in Organic Synthesis" (written by T.W. Greene; John Wiley & Sons, Inc.) and the like. The typical example includes a process in which R₂X₁H or HX₂-G-X₃H is, as described below, caused to act on the compound (XV) in the presence of an acid.

$$(XV)
\begin{array}{c}
 & \xrightarrow{R_2X_1H \text{ or} \\
 & HX_2-G-X_3H
\end{array}}$$

$$\xrightarrow{R_2X_1H \text{ or} \\
 & HX_2-G-X_3H
\end{array}}$$

$$(XXII)$$

wherein G, R_2 , X_1 , X_2 , X_3 and Z_2 have the same meanings as defined above. (b) The intended compound can be obtained by causing the nitrogen-containing compound (VI) or a salt thereof to act on the compound represented by the formula (XXII) in accordance with the following

wherein A, B, D, E₁, E₂, W, Y, Z₂, m and n have the same meanings as defined above.

The conversion from the compound (XXII) to the compound (Ib) can be conducted by treating the compound (XXII) with an organic or inorganic base and then causing the compound (VI) to act on the resulting compound, or causing the compound (VI) to act on the compound (XXII) in the presence of the inorganic or organic base. The conversion from the compound can be effected under the conditions similar to those shown in the conversion from the compound (XXII) to the compound (XXIII) in Process 1-(a). In this case, it is also possible to add an alkali iodide such as potassium iodide or sodium iodide at need. Incidentally, the compound (VI) can be synthesized by reacting the compound (V) with the compound (IIII) in a manner known per se in the art.

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reaction scheme:

Process 2:

Among the benzothiazine derivatives (I), each of compounds (Ic) in which Z is represented by a group:

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can be synthesized in any of the following processes.

(a) The intended compound can be obtained, in accordance with the following scheme, by converting the compound (XV) or (XXIII) to the compound (VIII) and then reacting the compound (VIII) with the compound represented by the formula (V):

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wherein A, B, D, E₁, E₂, W, W', Y, Z₂, m and n have the same meanings as defined above.

The conversion from the compound (XV) to the compound (VIII) can be effected under the conditions similar to those shown in the conversion from the compound (XXII) to the compound (XXIII) in Process 1-(a). Alternatively, the conversion from the compound (XXIII) to the compound (VIII) can be effected employing a method described in "Protective Groups in Organic Synthesis" (written by T.W. Greene; John Wiley & Sons, Inc.) and the like. For instance, the conversion to the intended compound (VIII) can be conducted by acid treatment of the compound (XXIII) when in Z₂, X₁ represents an oxygen atom or X₂ and X₃ both represent an oxygen atom, or by the treatment with mercury (II) chloride when also in

 Z_2 , X_1 represents a sulfur atom or X_2 and X_3 both represent a sulfur atom.

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The conversion from the compound (VIII) to the compound (Ic) can be effected under the conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1-(a).

(b) The intended compound can be obtained by the replacement of the group Z₂ of the compound (lb) with a carbonyl group in accordance with the following reaction scheme.

$$\begin{array}{c|c}
O \\
CH_2)_n \\
Y-(B)_m-D
\end{array}$$
(Ic)

wherein A, B, D, E₁, E₂, Y, Z₂, m and n have the same meanings as defined above.

The conversion from the compound (lb) to the compound (lc) can be effected under the conditions similar to those shown in the conversion from the compound (XXIII) to the compound (VIII) in Process 2(a).

Among the benzothiazine derivatives (I), each of the compounds (Ig) and (Ie) in which Z is represented by a group:

can be synthesized in accordance with any of the following processes. Where there is a group reactive to a hydroxylamine or a derivative thereof (VII) or a salt of the hydroxylamine or the derivative in a nitrogen-containing compound (V), it is desired to choose process (a).

(a) Each compound (Ig) can be obtained, in accordance with the following reaction scheme, by causing a hydroxylamine or a derivative thereof represented by the formula (VII) or a salt of the hydroxylamine or the derivative to act on the compound represented by the formula (VIII) and then causing the nitrogen-containing compound (V) to act further.

$$(CH_{2})_{n}$$

$$HN \qquad Y-(B)_{m}-D$$

$$E_{1} \qquad E_{2} \qquad (V)$$

$$O_{2} \qquad E_{1} \qquad E_{2}$$

$$(Ig)$$

wherein A, B, D, E₁, E₂, R₃, W, Y, m and n have the same meanings as defined above.

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The reaction between the compound (VIII) and the hydroxylamine or its derivative (VII) can be practiced, if necessary, in the presence of an organic base such as pyridine, triethylamine, collidine, DBU or sodium acetate or an inorganic base such as potassium carbonate or sodium hydroxide. The hydroxylamine or its derivative (VII) may also be used in the form of an organic acid salt or an inorganic acid salt.

The reaction is conducted at 0 °C to reflux temperature, preferably 0 °C to 100 °C optionally in a suitable solvent such as methanol, ethanol, propanol, tetrahydrofuran, dimethylformamide or dimethylsulfoxide

The conversion from the resulting compound (IX) to the compound (Ig) can be effected under the conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1(a).

(b) Each compound (le) can be obtained, in accordance with the following reaction scheme, by causing a hydroxylamine or a derivative thereof (VII) or a salt of the hydroxylamine or the derivative to act on the compound (Id):

(CH₂)_n

$$O_{2} = A-N \qquad Y-(B')_{m}-D$$

$$O_{3} = A-N \qquad Y-(B')_{m}-D$$

$$O_{4} = A-N \qquad Y-(B')_{m}-D$$

$$O_{5} = A-N \qquad Y-(B')_{m}-D$$

$$O_{6} = A-N \qquad Y-(B')_{m}-D$$

$$O_{7} = A-N \qquad Y-(B')_{m}-D$$

$$O_{8} = A-N \qquad Y-(B')_{m}-D$$

$$O_{1} = A-N \qquad Y-(B')_{m}-D$$

$$O_{2} = A-N \qquad Y-(B')_{m}-D$$

$$O_{3} = A-N \qquad Y-(B')_{m}-D$$

$$O_{4} = A-N \qquad Y-(B')_{m}-D$$

$$O_{5} = A-N \qquad Y-(B')_{m}-D$$

$$O_{6} = A-N \qquad Y-(B')_{m}-D$$

$$O_{7} = A-N \qquad Y-(B')_{m}-D$$

$$O_{8} = A-N \qquad Y-(B')_{m}-D$$

$$O_{1} = A-N \qquad Y-(B')_{m}-D$$

$$O_{2} = A-N \qquad Y-(B')_{m}-D$$

$$O_{3} = A-N \qquad Y-(B')_{m}-D$$

$$O_{4} = A-N \qquad Y-(B')_{m}-D$$

$$O_{5} = A-N \qquad Y-(B')_{m}-D$$

wherein, when Y represents CH, B' represents an oxygen atom, a sulfur atom, a sulfinyl group, a sulfonyl group, an alkelylene group, a substituted or unsubstituted hydroxymethylene group, a group -CHR $_5$ - in which R $_5$ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, or a substituted or unsubstituted cyclic or acyclic acetal group, when Y represents C = , B' represents a group of the following formula:

$$=$$
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in which the double bond is linked to Y, R_6 represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aralkyl group, or when Y represents a nitrogen atom, B' represents a carbonyl group, a sulfonyl group, an alkylene group, an alkenylene group, or a group -CHR₇- in which R_7 represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group; and A, D, E_1 , E_2 , R_3 , Y, m and n have the same meanings as defined above.

The conversion from the compound (Id) to the compound (Ie) can be effected under the conditions similar to those shown in the conversion from the compound (VIII) to the compound (IX) in Process 3(a).

45 Process 4:

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Among the benzothiazine derivatives (I), each of compounds (Ih) and (If) in which Z is represented by a group:

can be synthesized by any one of the following processes.

Incidentally, it is desired to select process (a) when there is a group reactive with a reducing agent in a nitrogen-containing compound (V).

(a) Each compound (lh) can be obtained by reducing the compound represented by the formula (VIII) to obtain the compound (X) and then causing the nitrogen-containing compound (V) to act on the resulting compound.

70 Reduction
$$OH$$

$$OH$$

$$O O$$

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wherein A, B, D, E₁, E₂, W, Y, m and n have the same meanings as defined above.

The conversion from the compound (VIII) to the compound (X) can be effected by treating the compound represented by the formula (VIII) with a reducing agent such as sodium borohydride, potassium borohydride, sodium cyanoborohydride or tri-n-tin hydride in an ordinarily-employed solvent at -78 °C to reflux temperature, preferably -20 °C to room temperature.

Further, the conversion from the compound (X) to the compound (Ih) can be effected under conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1(a).

(b) Each compound (If) can be obtained by reducing the compound (Id) in accordance with the following reaction scheme:

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(CH₂)_n
Y-(B')_m-D

Reduction

OH

(Id)

OH

$$CH_2$$
)_n
 CH_2

wherein A, B', D, E₁, E₂, Y, m and n have the same meanings as defined above.

The conversion from the compound (Id) to the compound (If) can be effected under conditions similar to those shown in the conversion from the compound (VIII) to the compound (X) in accordance with Process 4-(a).

Process 5:

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Among the benzothiazine derivatives (I), each compound (II) in which Z is represented by a group:

can be synthesized in accordance with the process which will be described hereinafter.

The intended compound can be obtained, in accordance with the following reaction scheme, by reacting the compound represented by the formula (XVI) with the compound represented by the formula (XVII), reacting the resulting compound with the compound represented by the formula (IIII) to obtain the compound represented by the formula (XXIV), and then causing a nitrogen-containing compound represented by the formula (V) to act on the compound (XXIV).

50 CH(OR₁)₃(XVI)
$$\longrightarrow$$
 NH \longrightarrow NH \longrightarrow NH \longrightarrow OR₁ \longrightarrow NH \longrightarrow OR₂ \longrightarrow O₂ \longrightarrow O₂ (XVII)

wherein A, B, D, E1, E2, R1, W, W', Y, m and n have the same meanings as defined above.

In the above reaction, the conversion from the compound (XVI) to the compound (XVII) can be effected by causing the compound (XVI) to act on the compound (XV) in the presence of p-toluenesulfonic acid, boron trifluoride ether complex, Amberlite 15 or the like.

Examples of the solvent usable in the above reaction may include methanol, ethanol, propanol and butanol. The reaction is conducted at -78 °C to reflux temperature.

The conversion from the compound (XVII) to the compound (II) can be effected under conditions similar to those shown in the conversion from the compound (XXII) to the compound (Ib) in Process 1(a).

Process 6:

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Among the benzothiazine derivatives (I), each compound (Ii) in which Z is represented by a group:

can be synthesized in accordance with the process which will be described hereinafter.

The compound represented by the formula (Ii) can be obtained, in accordance with the reaction scheme shown below, (1) by reducing the compound represented by the formula (XXV) to the compound represented by the formula (XX) and reacting the resulting compound with the compound represented by the formula (III) to obtain the compound (XII), or (2) by reacting the compound represented by the formula (X) with the compound represented by the formula (XI) to obtain the compound (XII), and then reacting the resulting compound (XII) with a nitrogen-containing compound represented by the formula (V).

In this case, it is desired to select a suitable process from the processes (1) and (2) according to the kind of group R_8 .

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wherein A, B, D, E₁, E₂, R₈, W, W', W'', Y, m and n have the same meanings as defined above.

In the above reaction, the conversion from the compound (XXV) to the compound (XX) can be conducted by treating, in the presence of a catalyst such as palladium-carbon or platinum, the compound (XXV) with hydrogen gas in an ordinarily-employed solvent at -78 °C to reflux temperature, preferably at room temperature. The conversion from the compound (XXI) to the compound (XIII) can be effected under conditions similar to those shown in the conversion from the compound (XXIII) to the compound (XXIIII) in Process 1(a).

The conversion from the compound (X) to the compound (XII) can be conducted by treating the compound (X) with an inorganic or organic base and then, causing the compound (XI) to act on the resulting compound, or causing the compound (XI) to act on the compound (X) in the presence of such a base.

Examples of group W" of the compound (XI), which is an eliminative substituent, include halogen atoms such as chlorine and bromine, alkylsulfonyloxy groups such as methanesulfonyloxy and arylsulfonyloxy groups such as p-toluenesulfonyloxy.

Further, exemplary inorganic or organic bases usable in the above reaction include sodium hydride, sodium bis(trimethylsilyl)amide, lithium diisopropylamide and potassium t-butoxide. Illustrative solvents usable in the present reaction include, tetrahydrofuran, dioxane, dimethylformamide, dimethylsulfoxide, N-methylpyrrolidone and toluene. The reaction may be conducted at -78°C to reflux temperature.

The conversion from the compound (XII) to the compound (Ii) can be effected under conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1(a).

Process 7:

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Among the benzothiazine derivatives (I), each compound (Ij) in which Z is represented by a group:

H C ✓ N

can be synthesized in accordance with the following process.

The compound represented by the formula (Ij) can be obtained, in accordance with the following reaction scheme, by subjecting the compound represented by formula (X) to dehydration to convert it into the compound represented by formula (XIII) and then causing a nitrogen-containing compound represented by formula (V) to act on the resulting compound.

20 Dehydration
$$O_2$$
 O_2 O_2 O_2 O_2 O_2 O_3 O_4

wherein A, B, D, E₁, E₂, W, Y, m and n have the same meanings as defined above.

In the above reaction, the conversion from the compound (X) to the compound (XIII) can be effected by causing methanesulfonyl chloride or p-toluenesulfonyl chloride and a base such as triethylamine, pyridine or collidine to act on the compound (X) in a solvent such as dichloromethane, chloroform or toluene and then treating the resulting product with the base described above or silica gel at room temperature to reflux temperature.

The conversion from the compound (XIII) to the compound (Ij) can be effected under conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1(a).

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Process 8:

Among the benzothiazine derivatives (I), each compound (Ik) in which Z is represented by a group:

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can be synthesized in accordance with the following process.

The compound represented by the formula (Ik) can be obtained, in accordance with the following reaction scheme, by reducing the compound represented by formula (XIII) to convert it into the compound represented by formula (XIV) and then reacting the resulting compound with a nitrogen-containing compound represented by formula (V).

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 $\begin{array}{c}
(CH_2)_n \\
+N \\
E_1 \\
E_2
\end{array}$

$$\begin{array}{c|c}
 & (CH_2)_n \\
 & & Y-(B)_m-D \\
\hline
 & (Ik)
\end{array}$$

wherein A, B, D, E₁, E₂, W, Y, m and n have the same meanings as defined above.

In the above reaction, the conversion from the compound (XIII) to the compound (XIV) can be effected under conditions similar to those shown in the conversion from the compound (XXV) to the compound (XX) in Process 6.

The conversion from the compound (XIV) to the compound (Ik) can be effected under conditions similar to those shown in the conversion from the compound (XXIII) to the compound (Ib) in Process 1(a).

If necessary, the compounds (I) of the present invention obtained according to the above-described processes can each be reacted with one of various acids to convert the compound to its salt. Then, the resulting salt can be purified by a method such as recrystallization or column chromatography.

Exemplary acids usable to convert the benzothiazine derivatives (I) to their salts include inorganic acids such as hydrochloric acid, nitric acid, sulfuric acid, phosphoric acid and hydrobromic acid; and organic acids such as maleic acid, fumaric acid, tartaric acid, lactic acid, citric acid, acetic acid, methanesulfonic acid, p-toluenesulfonic acid, adipic acid, palmitic acid and tannic acid.

As will be demonstrated later by test, the benzothiazine derivatives (I) and their salts according to the present invention, which can be obtained as described above, have a strong serotonin-2 blocking action and in addition, they have excellent selectivity of the serotonin-2 blocking action against α_1 blocking action. Further, as a result of a toxicity test, they have been found to feature high safety. The compounds according to the present invention can therefore be used as therapeutics for circulatory diseases such as ischemic heart diseases, cerebrovascular disturbance and peripheral circulatory disturbance.

When the benzothiazine derivative (I) according to this invention are used as drugs, they can be administered in an effective dose as they are. As an alternative, they can also be formulated into various preparation forms by known methods and then administered.

Exemplary preparation forms as drugs include orally administrable preparation forms such as tablets, powders, granules, capsules and syrups as well as parenterally administrable preparation forms such as injections and suppositories. Whichever preparation form is used, a known liquid or solid extender or carrier usable for the formulation of the preparation form can be employed.

Examples of such extender or carrier include polyvinylpyrrolidone, arabic gum, gelatin, sorbit, cyclodextrin, tragacanth gum, magnesium stearate, talc, polyethylene glycol, polyvinyl alcohol, silica, lactose, crystalline cellulose, sugar, starch, calcium phosphate, vegetable oil, carboxymethylcellulose, sodium laurylsulfate, water, ethanol, glycerin, mannitol, syrup, and the like.

When the compounds (I) according to the present invention are used as drugs, their dose varies depending on the administration purpose, the age, body weight and conditions of the patient to be administered, etc. In oral administration, the daily dose may generally be about 0.01-1,000 mg.

The present invention will next be described in further detail by the following examples and tests. But the present invention is not limited to the following examples and tests.

Example 1

by Synthesis of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide trimethylene acetal (Compound No. 1)

A mixture of 2.96 g (15 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide, 5.71 g (75 mmol) of trimethylene glycol, 285 mg (1.5 mmol) of p-toluenesulfonic acid monohydrate and 75 ml of toluene was refluxed for 30 hours in a container equipped with a Dean & Stark water separator.

The reaction mixture was cooled and then, 100 mL of a 0.02N aqueous solution of sodium hydroxide and ethyl acetate were added to the reaction mixture in this order, followed by fractionation. The resulting organic layer was washed with water and a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The oil so obtained was purified by chromatography on a silica gel column in which "No. 9385" (product of Merck & Co., the same silica gel was also used in the subsequent examples) was used as silica gel (eluent: ethyl acetate: hexane = 1:1), whereby 1.01 g of the title compound were obtained (yield: 26%).

Example 2

5 Synthesis of 3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-[(4'R,5'R)-dimethyl-1',3'-dioxolan] 1,1-dioxide (Compound No. 2)

A solution of 410 mg (2.08 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one1,1-dioxide, 517 mg (5.70 mmol) of (2R,3R)-2,3-butanediol and 38 mg (0.2 mmol) of p-toluenesulfonic acid monohydrate in 10 mt of benzene was refluxed for 20 hours in a container equipped with a Dean & Stark water separator.

Ethyl acetate was added to the reaction mixture. The resulting organic layer was washed with a saturated aqueous solution of sodium bicarbonate, water and saturated brine, dried over anhydrous magnesium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:ethyl acetate = 30:1), whereby 545 mg of the title compound were obtained (yield: 97%).

Example 3

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Synthesis of 3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-oxathiolan) 1,1-dioxide (Compound No. 3)

Under cooling and stirring, 615 μ t (5 mmol) of boron trifluoride ethyl ether complex were added to a solution of 986 mg (5 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide and 586 mg (7.5 mmol) of β -mercaptoethanol in 20 mt of methylene chloride. The reaction mixture was stirred for 27 hours at room temperature. An aqueous solution containing 691 mg (5 mmol) of potassium carbonate were thereafter added to the resulting mixture, followed by extraction with methylene chloride.

The resulting organic layer was washed with water and saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: chloroform), whereby 1.00 g of the title compound was obtained (yield 78%).

Example 4

Synthesis of 3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 4)

Under ice cooling and stirring, 250 µ1 (2 mmol) of boron trifluoride ethyl ether complex were added to a solution of 1.97 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide and 1.51 g (16 mmol) of 1,2-ethanedithiol in 38 m1 of methylene chloride. The reaction mixture was then stirred at room temperature.

Twenty eight hours later, another 250 µ1 (2 mmol) of boron trifluoride ethyl ether complex were added to the reaction mixture, followed by stirring for further 66 hours. The resulting reaction mixture was post-treated as in Example 3. The crude product so obtained was washed with chloroform, whereby 1.88 g of the title compound were obtained (yield: 69%).

Example 5

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Synthesis of 3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 5)

Under cooling and stirring, 710 mg (5 mmol) of boron trifluoride ethyl ether complex were added to a suspension of 1.97 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide and 1.62 g (15 mmol) of 1,3-propanedithiol in 50 mt of methylene chloride. The reaction mixture was stirred for 24 hours at room temperature. The crystals so precipitated were collected by filtration, followed by washing with methylene chloride.

The filtrate and the wash liquid were combined together, followed by washing with 100 mL of a 1% aqueous solution of potassium carbonate, water and saturated brine, drying over anhydrous sodium sulfate and concentration under reduced pressure. The crude crystals so obtained and the crystals collected by filtration were combined together and recrystallized from acetonitrile, whereby 2.55 g of the title compound were obtained (yield: 89%).

Example 6

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Synthesis of 4,4-bis(ethylthio)-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 6)

Under cooling and stirring, 50 μ t (0.4 mmol) of boron trifluoride ethyl ether complex were added to a solution of 197 mg (1 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide and 223 μ t (3 mmol) of ethanethiol in 5 mt of methylene chloride. The reaction mixture was stirred for 1 hour at room temperature. The post-treatment and purification were conducted as in Example 3, whereby 285 mg of the title compound were obtained (yield: 94%).

Example 7

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Synthesis of 4,4-dimethoxy-3,4-dihydro-2H-1,2-benzothiazine1,1-dioxide (Compound No. 7)

Under cooling and stirring, 0.75 m1 (6 mmol) of boron trifluoride ethyl ether complex was added to a solution of 985 mg (5 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one and 10 m1 (91.4 mmol) of methyl orthoformate in 10 m1 of methanol, followed by stirring at room temperature. After 85 hours, 0.75 m1 (6 mmol) of boron trifluoride ethyl ether complex was added further to the reaction mixture, followed by stirring for 20 hours.

Under ice cooling, the reaction mixture was added with 50 mt of a saturated aqueous solution of sodium bicarbonate and extracted with ethyl acetate. The organic layer was washed with saturated brine, dried over anhydrous magnesium sulfate and then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate: methylene chloride = 1:20), whereby 83 mg of the title compound were obtained (yield: 6.8%).

Example 8

Synthesis of 4-methoxy-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 8)

A mixture of 1.97 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide, 500 mg (2.6 mmol) of p-toluenesulfonic acid monohydrate, 20 ml of methyl orthoformate and 20 ml of methanol was refluxed for 3 hours. The solvent was distilled off and the residue was purified by chromatography on a silica gel column (eluent: methylene chloride:ethyl acetate = 40:1), whereby 2.06 g of the title compound were obtained (yield: 97%).

Example 9

Synthesis of 4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 9)

A suspension of 1.50 g (7.1 mmol) of Compound No. 8 and 300 mg of 10% palladium-carbon in 70 mt of ethanol was stirred vigorously for 24 hours under hydrogen gas atmosphere. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:ethyl acetate = 10:1), whereby 1.50 g of the title compound were obtained (yield: 99%).

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Example 10

Synthesis of 2-(2-chloroethyl)-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 10)

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To a suspension of 440 mg (11 mmol) of 60% sodium hydride in 20 mt of DMF, a solution of 2.41 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal in 10 mt of DMF was added under ice cooling and stirring. The reaction mixture was stirred for one hour at 0°C and then for further one hour at room temperature. Under ice cooling, a solution of 2.86 g (20 mmol) of 1-bromo-2-chloroethane in 10 mt of DMF was added to the reaction mixture, followed by stirring for 16 hours at room temperature.

The reaction mixture was concentrated under reduced pressure. To the residue, a 3:1 v/v mixed solvent of ethyl acetate and benzene was added. The organic layer was washed with a 5% aqueous solution of citric acid, water (twice) and a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:1), whereby 2.55 g of the title compound were obtained (yield: 84%).

Example 11

40 Synthesis of 2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 11)

To a suspension of 2.20 g (50 mmol) of 60% sodium hydride in 100 mt of DMF, a solution of 12.06 g (50 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal in 50 mt of DMF was added under ice cooling and stirring. The reaction mixture was stirred for 30 minutes at 0 °C and then for further one hour at room temperature. Under ice cooling, a solution of 16.95 g (150 mmol) of 1,3-dichloropropane in 50 mt of DMF was added to the reaction mixture, followed by stirring for 17 hours at room temperature. The post-treatment and purification were conducted as in Example 10, whereby 13.41 g of the title compound were obtained (yield: 84%).

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Example 12

Synthesis of 2-(4-chlorobutyl)-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 12)

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In a similar manner to Example 10 except that 1-bromo-2-chloroethane was replaced by the equimolar amount of 1,4-dichlorobutane, the reaction, post-treatment and purification were conducted, whereby 2.53 g of the title compound were obtained (yield: 76%).

Example 13

Synthesis of 2-(2-chloroethyl)-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 13)

As in Example 10, the reaction and post treatments were conducted using 547 mg (2 mmol) of Compound No. 4, 88 mg (2.2 mmol) of 60% sodium hydride, 574 mg (4 mmol) of 1-bromo-2-chloroethane and 12 m1 of DMF.

The crude crystals so obtained were recrystallized from ethyl acetate-hexane, whereby 533 mg of the title compound were obtained (yield: 79%).

Example 14

Synthesis of 2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 14)

As in Example 13, the reaction, post-treatment and purification were conducted employing 10.94 g (40 mmol) of Compound No. 4, 1.76 g (44 mmol) of 60% sodium hydride, 18.89 g (120 mmol) of 1-bromo-3-chloropropane and 160 mL of DMF, whereby 13.17 g of the title compound were obtained (yield: 87%).

Example 15

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Synthesis of 2-(2-chloroethyl)-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 15)

To a suspension of 66 mg (1.65 mmol) of 60% sodium hydride in 10 ml of DMF, a solution of 431 mg (1.5 mmol) of Compound No. 5 in 10 ml of DMF was added under ice cooling and stirring. The reaction mixture was stirred for 1.5 hours at room temperature and then for further 0.5 hour at 50 °C. Under ice cooling, a solution of 430 mg (3 mmol) of 1-bromo-2-chloroethane in 10 ml of DMF was added to the reaction mixture, followed by stirring for 16 hours at room temperature.

The reaction mixture was concentrated under reduced pressure. To the residue, 50 mt of a half-saturated aqueous solution of potassium carbonate were added, followed by extraction with chloroform. The organic layer was washed with saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was subjected to chromatography on a silica gel column (eluent: chloroform). The crude crystals so obtained were recrystallized from ethyl acetate-hexane, whereby 403 mg of the title compound were obtained (yield: 77%).

Example 16

Synthesis of 2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 16)

As in Example 15, the reaction and post-treatment were conducted using 1.15 g (4 mmol) of Compound No. 5, 176 mg (4.4 mmol) of 60% sodium hydride, 1.36 g (12 mmol) of 1,3-dichloropropane and 50 mL of DMF.

The residue was subjected to chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:2). The crude products so obtained were recrystallized from ethyl acetate-hexane, whereby 1.00 g of the title compound were obtained (yield: 69%).

so Example 17

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Synthesis of 2-(3-bromopropyl)-4,4-dimethoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 17)

A suspension of 83 mg (0.34 mmol) of Compound No. 7, 683 mg (3.4 mmol) of 1,3-dibromopropane and 97 mg (0.70 mmol) of potassium carbonate in 5 mt of acetone was refluxed for 4 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The oil so obtained was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:3), whereby 112 mg of

the title compound were obtained (yield: 90%).

Example 18

Synthesis of 2-(3-bromopropyl)-4-methoxy-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 18)

A suspension of 527 mg (2.5 mmol) of Compound No. 8, 2.51 g (12.5 mmol) of 1,3-dibromopropane and 690 mg (5 mmol) of potassium carbonate in 37.5 mL of acetone was refluxed for 2 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:4), whereby 706 mg of the title compound were obtained (yield: 85%).

Example 19

15 Synthesis of 2-(3-bromopropyl)-4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 19)

A suspension of 213 mg (1 mmol) of Compound No. 9, 1.00 g (5 mmol) of 1,3-dibromopropane and potassium carbonate in 10 mt of acetone was refluxed for 4 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:3), whereby 294 mg of the title compound were obtained (yield: 88%).

Example 20

Synthesis of 2-(3-chloropropyl)-4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 20)

A suspension of 852 mg (4 mmol) of Compound No. 9, 942 mg (6 mmol) of 1-bromo-3-chloropropane and 1.10 g (8 mmol) of potassium carbonate in 40 mt of acetone was refluxed for 10 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:2), whereby 1.11 g of the title compound were obtained (yield: 95%).

35 Example 21

Synthesis of 2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide (Compound No. 21)

A mixture of 3.18 g (10 mmol) of Compound No. 11, 40 mL of 3N hydrochloric acid and 40 mL of methanol was refluxed for 5 hours. The reaction mixture was concentrated under reduced pressure. To the residue, water was added, followed by extraction with methylene chloride. The organic layer was washed with a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 3:2), whereby 2.66 g of the title compound were obtained (yield: 97%).

(Another process)

In a similar manner to Example 10 except that 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal and 1-bromo-2-chloroethane were replaced by 1.97 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one and 4.72 g (30 mmol) of 1-bromo-3-chloropropane, respectively, the reaction, post-treatment and purification were conducted, whereby 721 mg of the title compound were obtained (yield: 26%).

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Example 22

Synthesis of 2-(3-chloropropyl)-4-hydroxyimino-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 22)

A mixture of 1.64 g (6 mmol) of Compound No. 21, 738 mg (9 mmol) of sodium acetate, 625 mg (9 mmol) of hydroxylamine hydrochloride and 50 m1 of methanol was refluxed for 40 hours.

The reaction mixture was concentrated under reduced pressure. To the residue, a half-saturated aqueous solution of potassium carbonate was added, followed by extraction with chloroform. The organic layer was washed with a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:2), whereby 1.51 g of the title compound were obtained (yield: 87%).

Example 23

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Synthesis of 2-(3-chloropropyl)-4-hydroxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 23)

Under ice cooling and stirring, 378 mg (10 mmol) of sodium borohydride were added in portions to a solution of 1.09 g (4 mmol) of Compound No. 21 in 40 mL of ethanol. The reaction mixture was stirred for 2 hours at 0 °C and then for 18 hours at room temperature.

The reaction mixture was concentrated under reduced pressure. To the residue, water was added, followed by extraction with methylene chloride. The organic layer was washed with a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:1), whereby 1.03 g of the title compound were obtained (yield: 93%).

(Another process)

A mixture of 398 mg (2 mmol) of 4-hydroxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide, 471 mg (3 mmol) of 1-bromo-3-chloropropane, 552 mg (4 mmol) of potassium carbonate and 20 mL of acetone was refluxed for 10 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:ethyl acetate = 20:1), whereby 514 mg of the title compound were obtained (yield: 93%).

35 Example 24

Synthesis of 2-(3-chloropropyl)-4-ethoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 24)

Under ice cooling and stirring, 23 mg (0.57 mmol) of 60% sodium hydride and 0.08 m1 (0.94 mmol) of ethyl iodide were successively added to a solution of 130 mg (0.47 mmol) of Compound No. 23 in 2 m1 of DMF. The reaction mixture was stirred for one hour under ice cooling and then, for further one hour at room temperature.

Ethyl acetate was added to the reaction mixture. The resulting mixture was washed with an aqueous solution of 10% citric acid (twice), water and a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:3), whereby 125 mg of the title compound were obtained (yield: 87%).

Example 25

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Synthesis of 4-benzyloxy-2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 25)

Under ice cooling, 24 mg (0.60 mmol) of 60% sodium hydride and 0.12 mt (1.0 mmol) of benzyl bromide were added successively to a solution of 138 mg (0.50 mmol) of Compound No. 23 in 2 mt of DMF. The reaction mixture was stirred for one hour under ice cooling and for further one hour at room temperature. The reaction mixture was post-treated as in Example 24. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:4), whereby 139 mg of the title

compound were obtained (yield: 76%).

Example 26

5 Synthesis of 2-(3-chloropropyl)-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 26)

Under ice cooling and stirring, 1.2 m1 (15 mmol) of methanesulfonyl chloride were added to a solution of 1.44 g (5.2 mmol) of Compound No. 23, 3.03 g (30 mmol) of triethylamine in 52 m1 of methylene chloride, followed by stirring at 0 ° C for one hour.

Ethyl acetate was added to the reaction mixture. The organic layer was washed with 0.5N hydrochloric acid, water, a saturated aqueous solution of sodium bicarbonate and saturated brine, dried over anhydrous sodium sulfate and then concentrated under reduced pressure. The residue was dissolved in 100 mt of methylene chloride and 50 mt of ethyl acetate, followed by the addition of 50 g of silica gel. After the resulting mixture was stirred at room temperature for 17 hours, the reaction mixture was filtered. The solid matter so obtained was extracted with ethyl acetate. The filtrate and the extract were combined together, followed by washing with a saturated aqueous solution of sodium bicarbonate and saturated brine, drying over anhydrous sodium sulfate and concentration under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:ethyl acetate = 20:1), whereby 937 mg of the title compound were obtained (yield: 70%).

Example 27

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Synthesis of 2-(3-chloropropyl)-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 27)

To a solution of 82 mg (0.32 mmol) of Compound No. 26 and three drops of acetic acid in 10 mL of ethanol, 20 mg of 10% palladium-carbon were added, followed by stirring under a hydrogen gas stream for 72 hours. The reaction mixture was filtered. The filtrate was concentrated under reduced pressure. The residue so obtained was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:3), whereby 73 mg of the title compound were obtained (yield: 88%).

Example 28

Synthesis of 2-[3-(4-phenylpiperazin-1-yl)propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 28)

A suspension of 1.27 g (4 mmol) of Compound No. 11, 2.60 g (16 mmol) of 1-phenylpiperazine and 6.00 g (40 mmol) of sodium iodide in 95 mt of DMF was stirred at 80°C for 16 hours.

To the reaction mixture, 400 mt of a 3:1 v/v mixed solvent of ethyl acetate and benzene were added. The organic layer was washed with a half-saturated aqueous solution of potassium carbonate, water and a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was subjected to chromatography on a silica gel column (eluent: ethyl acetate). The crude crystals so obtained were recrystallized from 2-propanol-isopropyl ether, whereby 1.57 g of the title compound were obtained (yield: 88%).

45 Example 29

Synthesis of 2-[2-[4-(4-fluorophenyl)piperazin-1-yl]ethyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 29)

A suspension of 607 mg (2 mmol) of Compound No. 10, 721 mg (4 mmol) of 1-(4-fluorophenyl)-piperazine and 600 mg (4 mmol) of sodium iodide in 30 mt of acetonitrile was refluxed for 16 hours.

The reaction mixture was concentrated under reduced pressure. To the residue, ethyl acetate was added. The organic layer was washed with a half-saturated aqueous solution of potassium carbonate, water and a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:1 → ethyl acetate), whereby 443 mg of the title compound were obtained (yield: 49%).

Example 30

Synthesis of 2-[4-[4-(4-fluorophenyl)piperazin-1-yl]butyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 30)

In a similar manner to Example 29 except that Compound No. 10 was replaced by 663 mg (2 mmol) of Compound No. 12, the reaction, post-treatment and purification were conducted, whereby 697 mg of the title compound were obtained (yield: 73%).

Example 31

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Synthesis of 2-[3-(4-diphenylmethylpiperazin-1-yl)propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 31)

A suspension of 636 mg (2 mmol) of Compound No. 11, 606 mg (2.4 mmol) of 1-diphenylmethyl-piperazine and 332 mg (2.4 mmol) of potassium carbonate in 30 mt of acetonitrile was refluxed for 40 hours. The reaction mixture was post-treated as in Example 22. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 3:97). The crude crystals so obtained were recrystallized from ethanol, whereby 760 mg of the title compound were obtained (yield: 71%).

Example 32

Synthesis of 2-[3-(4-phenylpiperidino)propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 32)

A suspension of 636 mg (2 mmol) of Compound No. 11, 484 mg (3 mmol) of 4-phenylpiperidine, 415 mg (3 mmol) of potassium carbonate and 600 mg (4 mmol) of sodium iodide in 30 mt of acetonitrile was refluxed for 13 hours.

The post-treatment and purification were conducted as in Example 29, whereby 693 mg of the title compound were obtained (yield: 78%).

Example 33

Synthesis of 2-[3-[4-(3-methoxyphenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-35 dioxide ethylene acetal (Compound No. 33)

A suspension of 318 mg (1 mmol) of Compound No. 11, 288 mg (1.5 mmol) of 1-(3-methoxyphenyl)-piperazine, 207 mg (1.5 mmol) of potassium carbonate and 300 mg (2 mmol) of sodium iodide in 15 mL of acetonitrile was refluxed for 17 hours. The reaction mixture was post-treated as in Example 31. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 3:97). The crude crystals so obtained were recrystallized from ethyl acetate-hexane, whereby 326 mg of the title compound were obtained (yield: 69%).

Example 34

Synthesis of 2-[3-[4-(4-methoxyphenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound 34)

A suspension of 318 mg (1 mmol) of Compound No. 11, 264 mg (1 mmol) of 1-(4-methoxyphenyl)-piperazine dihydrochloride, 420 mg (5 mmol) of sodium bicarbonate and 300 mg (2 mmol) of sodium iodide in 15 m1 of acetonitrile was refluxed for 20 hours. The reaction mixture was post-treated as in Example 31. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 1:99), whereby 398 mg of the title compound were obtained (yield: 84%).

Example 35

Synthesis of 2-[3-[4-(2-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound 35)

A suspension of 318 mg (1 mmol) of Compound No. 11, 217 mg (1 mmol) of 1-(2-fluorophenyl)-piperazine hydrochloride, 336 mg (4 mmol) of sodium bicarbonate and 300 mg (2 mmol) of sodium iodide in 15 mL of acetonitrile was refluxed for 18 hours. The reaction mixture was post-treated as in Example 31. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 1:99). The crude crystals so obtained were recrystallized from ethyl acetate-hexane-ethyl ether, whereby 233 mg of the title compound were obtained (yield: 50%).

Example 36

75 Synthesis of 2-[3-[4-(3-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 36)

In a similar manner to Example 35 except that 1-(2-fluorophenyl)piperazine hydrochloride was replaced by 261 mg (1 mmol) of 1-(3-fluorophenyl)piperazine hydrobromide, the reaction and post-treatment were conducted. The residue was purified by chromatography on a silica gel column (eluent: chloroform), whereby 369 mg of the title compound were obtained (yield: 80%).

Example 37

Synthesis of 2-[3-[4-(4-hydroxyphenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 37)

In a similar manner to Example 35 except that 1-(2-fluorophenyl)piperazine hydrochloride was replaced by 259 mg (1 mmol) of 1-(4-hydroxyphenyl)piperazine hydrobromide, the reaction and post-treatment were conducted. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 3:97), whereby 286 mg of the title compound were obtained (yield: 62%).

Example 38

35 Synthesis of 2-[3-[4-(4-chlorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 38)

In a similar manner to Example 35 except that 1-(2-fluorophenyl)piperazine hydrochloride was replaced by 233 mg (1 mmol) of 1-(4-chlorophenyl)piperazine hydrochloride, the reaction and post-treatment were conducted. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 1:99), whereby 434 mg of the title compound were obtained (yield: 90%).

Example 39

Synthesis of 2-[3-[4-(4-methanesulfonamidophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound 39)

A suspension of 159 mg (0.5 mmol) of Compound No. 11, 146 mg (0.5 mmol) of 4-(4-methanesul-fonamidophenyl)piperazine hydrochloride, 168 mg (2 mmol) of sodium bicarbonate and 150 mg (1 mmol) of sodium iodide in 15 m² of acetonitrile was refluxed for 16 hours. The reaction mixture was post-treated in a similar manner to Example 22. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 1:99 → 2:98 → 3:97), whereby 133 mg of the title compound were obtained (yield: 50%).

Example 40

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Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-3,4-dihydro-2H-1,2-bensothiazine-4-spiro-2'-(1',3'-dioxolan) 1,1-dioxide (Compound No. 40)

A suspension of 795 mg (2.5 mmol) of Compound No. 11, 609 mg (2.5 mmol) of 4-(4-fluorobenzoyl)-piperidine hydrochloride, 804 mg (10 mmol) of sodium bicarbonate and 750 mg (5 mmol) of sodium iodide in 25 mL of acetonitrile was refluxed for 26 hours. The post-treatment and purification were conducted as in Example 31, whereby 1.02 g of the title compound were obtained (yield: 84%).

Example 41

Synthesis of 2-[3-[4-(2,4-difluorobenzoyl)piperidino]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 41)

In a similar manner to Example 39 except that 4-(4-methanesulfonamidophenyl)piperazine hydrochloride was replaced by 157 mg (0.6 mmol) of 4-(2,4-difluorobenzoyl)piperidine hydrochloride, the reaction and the post treatments were conducted. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 15:1), whereby 171 mg of the title compound were obtained (yield: 68%).

Example 42

Synthesis of 2-[2-[4-(6-fluoro-1,2-benzisoxazol-3-yl)piperidino]ethyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1.1-dioxide ethylene acetal (Compound No. 42)

A suspension of 304 mg (1 mmol) of Compound No. 10, 264 mg (1.2 mmol) of 4-(6-fluoro-1,2-benzisoxazol-3-yl)piperidine, 168 mg (2 mmol) of sodium bicarbonate and 300 mg (2 mmol) of sodium iodide in 20 m1 of acetonitrile was refluxed for 18 hours.

The reaction mixture was post-treated as in Example 22. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride: methanol = 30: 1), whereby 270 mg of the title compound were obtained (yield: 55%).

Example 43

Synthesis of 2-[3-[4-(6-fluoro-1,2-benzisoxazol-3-yl)piperidino]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 43)

A suspension of 95 mg (0.3 mmol) of Compound No. 11, 72 mg (0.3 mmol) of 4-(6-fluoro-1,2-40 benzisoxazol-3-yl)piperidine, 101 mg (1.2 mmol) of sodium bicarbonate and 90 mg (0.6 mmol) of sodium iodide in 7.5 ml of acetonitrile was refluxed for 22 hours. The reaction mixture was post-treated and purified as in Example 42, whereby 128 mg of the title compound were obtained (yield: 85%).

Example 44

Synthesis of 2-[3-[4-(1,2-benzisothiazol-3-yl]piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 44)

A suspension of 318 mg (1 mmol) of Compound No. 11, 329 mg (1.5 mmol) of 1-(1,2-benzisothiazol-3yl)piperazine, 207 mg (1.5 mmol) of potassium carbonate and 300 mg (2 mmol) of sodium iodide in 15 mt
of acetonitrile was refluxed for 15 ours. The reaction mixture was post-treated as in Example 31. The
residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 1:99),
whereby 501 mg of the title compound were obtained (yield: 100%).

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Example 45

Synthesis of 2-[3-[4-(6-fluoro-1H-indazol-3-yl)piperidino]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 45)

A suspension of 159 mg (0.5 mmol) of Compound No. 11, 164 mg (0.75 mmol) of 4-(6-fluoro-1H-indazol-3-yl)piperidin, 104 mg (0.75 mmol) of potassium carbonate and 150 mg (1 mmol) of sodium iodide in 10 mt of acetonitrile was refluxed for 15 hours. The reaction mixture was post-treated as in Example 22. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 1:19). The crude crystals so obtained were recrystallised from chloroform-ethyl ether, whereby 204 mg of the title compound were obtained (yield: 82%).

Example 46

5 Synthesis of 2-[3-[4-(2-pyridyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 46)

In a similar manner to Example 44 except that 1-(1,2-benzisothiazol-3-yl)piperazine was replaced by 245 mg (1.5 mmol) of 1-(2-pyridyl)piperazine, the reaction and post-treatment were conducted. The residue so obtained was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 2:98), whereby 336 mg of the title compound were obtained (yield: 76%).

Example 47

 Synthesis of 2-[3-[4-(2-pyrimidyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 47)

In a similar manner to Example 34 except that 1-(4-methoxyphenyl)piperazine dihydrochloride was replaced by 237 mg (1 mmol) of 1-(2-pyrimidyl)piperazine dihydrochloride, the reaction and post-treatment were conducted. The residue so obtained was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 2:98), whereby 390 mg of the title compound were obtained (yield: 88%).

Example 48

Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 48)

A suspension of 350 mg (1 mmol) of Compound No. 14, 244 mg (1 mmol) of 4-(4-fluorobenzoyl)-piperidine hydrochloride, 336 mg (4 mmol) of sodium bicarbonate and 300 mg (2 mmol) of sodium iodide in 15 mt of acetonitrile was refluxed for 17 hours. The reaction mixture was post-treated as in Example 22. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 97:3), whereby 450 mg of the title compound were obtained (yield: 86%).

Example 49

Synthesis of 2-[3-[4-(6-fluoro-1H-indazol-3-yl)piperidino]propyl]-3,4-dihydro-2H-1,2- benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 49)

In a similar manner to Example 48 except that 4-(4-fluorobenzoyl)piperidine hydrochloride was replaced by 219 mg (1 mmol) of 4-(6-fluoro-1H-indazol-3-yl)piperidine, the reaction, post-treatment and purification were conducted, whereby 240 mg of the title compound were obtained (yield: 45%).

Example 50

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Synthesis of 2-[3-[4-(4-hydroxyphenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 50)

In a similar manner to Example 48 except that 4-(4-fluorobenzoyl)piperidine hydrochloride was replaced by 259 mg (1 mmol) of 1-(4-hydroxyphenyl)piperazine hydrobromide, the reaction and post-treatment were conducted. The residue was subjected to chromatography on a silica gel column (eluent: methylene chloride: methanol = 97:3 \rightarrow 19:1). The crude crystals so obtained were recrystallized from acetonitrile-isopropyl ether, whereby 270 mg of the title compound were obtained (yield: 55%).

Example 51

Synthesis of 2-[2-[4-(4-fluorophenyl)piperazin-1-yl]ethyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro- 2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 51)

A suspension of 336 mg (1 mmol) of Compound No. 13, 276 mg (1.5 mmol) of 1-(4-fluorophenyl)-piperazine, 504 mg (6 mmol) of sodium bicarbonate and 300 mg (2 mmol) of sodium iodide in 15 mL of acetonitrile was refluxed for 44 hours. The reaction mixture was post-treated as in Example 31. The residue was subjected to chromatography on a silica gel column (eluent: ethyl acetate:hexane = 1:1). The crude crystals so obtained were recrystallized from ethyl acetate-hexane, whereby 315 mg of the title compound were obtained (yield: 66%).

Example 52

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Synthesis of 2-[2-[4-(4-hydroxyphenyl)piperazin-1-yl]ethyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 52)

A suspension of 168 mg (0.5 mmol) of Compound No. 13, 130 mg (0.5 mmol) of 1-(4-hydroxyphenyl)-piperazine hydrobromide, 168 mg (2 mmol) of sodium bicarbonate and 150 mg (1 mmol) of sodium iodide in 15 mL of acetonitrile was refluxed for 20 hours. The reaction mixture was filtered and the filtrate was concentrated under reduced pressure. The residue was subjected to chromatography on a silica gel column (eluent: methanol:chloroform = 3:97). The crude crystals so obtained were recrystallized from ethyl acetate-hexane, whereby 78 mg of the title compound were obtained (yield: 33%).

Example 53

Synthesis of 2-[3-[4-(4-hydroxyphenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 53)

As in Example 52, the reaction and post-treatment were conducted employing 364 mg (1 mmol) of Compound No. 16, 259 mg (1 mmol) of 1-(4-hydroxyphenyl)piperazine hydrobromide, 336 mg (4 mmol) of sodium bicarbonate, 300 mg (2 mmol) of sodium iodide and 20 mL of acetonitrile. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 3:97), whereby 337 mg of the title compound were obtained (yield: 67%).

Example 54

Synthesis of 2-[2-[4-(4-hydroxyphenyl)piperazin-1-yl]ethyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 54)

In a similar manner to Example 53 except that Compound No. 16 was replaced by 350 mg (1 mmol) of Compound No. 15, the reaction and post-treatment were conducted. The residue was subjected to chromatography on a silica gel column (eluent: methanol: chloroform = 3:97). The crude crystals so obtained were recrystallized from acetonitrile-isopropyl ether, whereby 100 mg of the title compound were obtained (yield: 20%).

Example 55

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Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl)propyl]-4,4-dimethoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 55)

A suspension of 112 mg (0.3 mmol) of Compound No. 17, 81 mg (0.45 mmol) of 1-(4-fluorophenyl)-piperazine and 83 mg (0.6 mmol) of potassium carbonate in 10 mL of 1,4-dioxane was stirred at 90 °C for 15 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methanol:methylene chloride = 1:30), whereby 146 mg of the title compound were obtained (yield: 99%).

Example 56

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-4-methoxy-2H-1,2-benzothiazine 1,1-dioxide

A suspension of 664 mg (2 mmol) of Compound No. 18, 432 mg (2.4 mmol) of 1-(4-fluorophenyl)-piperazine and 552 mg (4 mmol) of potassium carbonate in 30 mt of 1,4-dioxane was refluxed for 13 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 40:1), whereby 785 mg of the title compound were obtained (yield: 91%).

Example 57

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 57)

A suspension of 250 mg (0.74 mmol) of Compound No. 19, 170 mg (0.90 mmol) of 1-(4-fluorophenyl)-piperazine and 207 mg (1.5 mmol) of potassium carbonate in 10 mL of 1,4-dioxane was refluxed for 6 hours. The reaction mixture was filtered and the filtrate was then concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 30:1), whereby 303 mg of the title compound were obtained (yield: 94%).

Example 58

Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 58)

A suspension of 145 mg (0.5 mmol) of Compound No. 20, 182 mg (0.75 mmol) of 4-(4-fluorobenzoyl)-piperidine hydrochloride, 168 mg (2 mmol) of sodium bicarbonate and 150 mg (1 mmol) of sodium iodide in 10 mL of acetonitrile was refluxed for 20 hours. The reaction mixture was post-treated as in Example 31. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride: methanol = 20:1), whereby 220 mg of the title compound were obtained (yield: 95%).

45 Example 59

Synthesis of 4-ethoxy-2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 59)

A suspension of 152 mg (0.5 mmol) of Compound No. 24, 135 mg (0.75 mmol) of 1-(4-fluorophenyl)piperazine, 84 mg (1.0 mmol) of sodium bicarbonate and 150 mg (1 mmol) of sodium iodide in 10 mL of
acetonitrile was refluxed for 19 hours. The reaction mixture was post-treated as in Example 31. The residue
was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 30:1),
whereby 190 mg of the title compound were obtained (yield: 85%).

Example 60

Synthesis of 2-[3-[4-(4-hydroxyphenyl)piperazin-1-yl]propyl]-4-methoxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 60)

A suspension of 145 mg (0.5 mmol) of Compound No. 20, 155 mg (0.6 mmol) of 1-(4-hydroxyphenyl)-piperazine hydrobromide, 168 mg (2 mmol) of sodium bicarbonate and 150 mg (1 mmol) of sodium iodide in 10 mt of acetonitrile was refluxed for 20 hours. The reaction mixture was post-treated as in Example 31. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 15:1), whereby 70 mg of the title compound were obtained (yield: 32%).

Example 61

Synthesis of 4-benzyloxy-2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 61)

A suspension of 109 mg (0.3 mmol) of Compound No. 25, 81 mg (0.45 mmol) of 1-(4-fluorophenyl)-piperazine, 50 mg (0.6 mmol) of sodium bicarbonate and 90 mg (0.6 mmol) of sodium iodide in 6 ml of acetonitrile was refluxed for 18 hours. The reaction mixture was post-treated as in Example 31. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 40:1), whereby 120 mg of the title compound were obtained (yield: 78%).

Example 62

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal (Compound No. 62)

Under ice cooling and stirring, a solution of 2.42 g (10 mmol) of 3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide ethylene acetal in 20 mt of DMF was added to a suspension of 480 mg (12 mmol) of 60% sodium hydride in 60 mt of DMF. The reaction mixture was stirred for one hour at 0 °C and for further one hour at room temperature.

The reaction mixture was thereafter cooled to 0 °C, to which a solution of 3.85 g (15 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine in 20 m± of DMF was added. The resulting mixture was stirred at room temperature for 58 hours. The reaction mixture was post-treated as in Example 28. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 3:1 → ethyl acetate), whereby 4.30 g of the title compound were obtained (yield: 93%).

Example 63

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro- 2'-[-(4'R,5'R)-dimethyl-1',3'-dioxolan] 1,1-dioxide (Compound No. 63)

As in Example 62, the reaction was conducted using 24 mg (0.6 mmol) of 60% sodium hydride, 135 mg (0.5 mmol) of Compound No. 2, 192 mg (0.75 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine and 6 mt of DMF. The reaction mixture was post-treated as in Example 22. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride: methanol = 20: 1), whereby 176 mg of the title compound were obtained (yield: 72%).

Example 64

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Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dioxane) 1,1-dioxide (Compound No. 64)

As in Example 62, the reaction, post-treatment and purification were conducted employing 185 mg (4.63 mmol) of 60% sodium hydride, 987 mg (3.86 mmol) of Compound No. 1, 1.49 g (5.79 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine and 50 mt of DMF, whereby 1.33 g of the title compound were obtained (yield: 72%).

Example 65

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro- 2'-(1',3'-oxathiolan) 1,1-dioxide (Compound No. 65)

The reaction, post-treatment and purification were conducted as in Example 62 by employing 96 mg (2.4 mmol) of 60% sodium hydride, 515 mg (2 mmol) of Compound No. 3, 770 mg (3 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine and 25 m1 of DMF, whereby 635 mg of the title compound were obtained (yield: 66%).

Example 66

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Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithiolan) 1,1-dioxide (Compound No. 66)

The reaction, post-treatment and purification were conducted as in Example 62 by employing 88 mg (2.2 mmol) of 60% sodium hydride, 547 mg (2 mmol) of Compound No. 4, 770 mg (3 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine and 25 mt of DMF, whereby 950 mg of the title compound were obtained (yield: 96%).

Example 67

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 67)

As in Example 62, the reaction was conducted using 44 mg (1.1 mmol) of 60%-sodium hydride, 287 mg (1 mmol) of Compound No. 5, 385 mg (1.5 mmol) of 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine and 20 mt of DMF.

The reaction mixture was concentrated under reduced pressure. A half-saturated aqueous solution of potassium carbonate was added to the residue, followed by extraction with ethyl acetate. The organic layer was washed with saturated brine, dried over anhydrous sodium sulfate and concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: chloroform), whereby 442 mg of the title compound were obtained (yield: 87%).

35 Example 68

Synthesis of 2-[2-[4-(4-fluorophenyl)piperazin-1-yl]ethyl]-3,4-dihydro-2H-1,2-benzothiazine-4-spiro-2'-(1',3'-dithian) 1,1-dioxide (Compound No. 68)

40 In a similar manner to Example 67 except that 1-(3-chloropropyl)-4-(4-fluorophenyl)piperazine was replaced by 364 mg (1.5 mmol) of 1-(2-chloroethyl)-4-(4-fluorophenyl)piperazine, the reaction was conducted

As in Example 51, the post-treatments and purification were conducted, whereby 247 mg of the title compound were obtained (yield: 50%).

Example 69

Synthesis of 4,4-bis(ethylthio)-2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 69)

In a similar manner to Example 65 except that Compound No. 3 was replaced by 607 mg (2 mmol) of Compound No. 6, the reaction, post-treatment and purification were conducted, whereby 814 mg of the title compound were obtained (yield: 78%).

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Example 70

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide (Compound No. 70)

A mixture of 923 mg (2 mmol) of Compound No. 62, 25 mt of 3N hydrochloric acid and 25 mt of methanol was refluxed for 2 hours. The reaction mixture was concentrated under reduced pressure. To the residue, 200 mt of a half-saturated aqueous solution of potassium carbonate were added, followed by extraction with chloroform. The organic layer was washed with a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The residue was purified by chromatography on a silica gel column (eluent: ethyl acetate:hexane = 2:1), whereby 772 mg of the title compound were obtained (yield: 92%).

Example 71

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Synthesis of 2-[3-(4-phenylpipierazin-1-yl)propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide (Compound No. 71)

In a similar manner to Example 70 except that Compound No. 62 was replaced by 887 mg (2 mmol) of Compound No. 28, the reaction, post-treatment and purification were conducted, whereby 719 mg of the title compound were obtained (yield: 90%).

Example 72

25 Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-3,4-dihydro-2H-1,2-benzothiazin-4-one 1,1-dioxide (Compound No. 72)

A mixture of 600 mg (1.2 mmol) of Compound No. 40, 5 mt of 3N hydrochloric acid and 5 mt of methanol was refluxed for 1.5 hours. The reaction mixture was post-treated as in Example 70. The residue was purified by chromatography on a silica gel column (eluent: methylene chloride:methanol = 30:1), whereby 483 mg of the title compound were obtained (yield: 90%).

Example 73

35 Synthesis of 4-hydroxyimino-2-[3-(4-phenylpiperazin-1-yl)propyl]-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 73)

A suspension of 230 mg (0.797 mmol) of Compound No. 22, 517 mg (3.19 mmol) of 1-phenylpiperazine and 1.19 g (7.97 mmol) of sodium iodide in 20 mt of DMF was stirred at 80 °C for 16 hours. The post-treatment and purification were conducted as in Example 62, whereby 239 mg of the title compound were obtained (yield: 72%).

Example 74

45 Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-4-hydroxyimino-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 74)

A suspension of 722 mg (2.5 mmol) of Compound No. 22, 609 mg (2.5 mmol) of 4-(4-fluorobenzoyl)-piperidine hydrochloride, 840 mg (10 mmol) of sodium bicarbonate and 749 mg (5 mmol) of sodium iodide in 50 mL of acetonitrile was refluxed for 24 hours.

The reaction mixture was concentrated under reduced pressure. To the residue, a half-saturated aqueous solution of potassium carbonate was added, followed by extraction with dichloromethane and ethyl acetate. The organic layer was washed with a saturated brine, dried over anhydrous sodium sulfate and then, concentrated under reduced pressure. The solid matters so obtained were washed with methanol and then, recrystallized from acetonitrile, whereby 533 mg of the title compound were obtained (yield: 46%).

Example 75

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Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-4-hydroxyimino-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 75)

A solution of 1.25 g (3 mmol) of Compound No. 70, 417 mg (6 mmol) of hydroxylamine hydrochloride and 492 mg (6 mmol) of sodium acetate in 30 mt of methanol was refluxed for 3 hours. The reaction mixture was post-treated as in Example 22. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 3:97), whereby 1.27 g of the title compound were obtained (yield: 98%).

Example 76

Synthesis of 2-[3-[4-(4-fluorobenzoyl)piperidino]propyl]-4-hydroxy-3,4-dihydro-2H-1,2-benzothiazine 1,115 dioxide (Compound No. 76)

In a similar manner to Example 74 except that Compound No. 22 was replaced by 689 mg (2.5 mmol) of Compound No. 23, the reaction was conducted. The reaction mixture was post-treated as in Example 31. The residue was purified by chromatography on a silica gel column (eluent: methanol:chloroform = 7.5:92.5), whereby 818 mg of the title compound were obtained (yield: 92%).

Example 77

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-4-hydroxy-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 77)

Under ice cooling and stirring, 300 mg (7.9 mmol) of sodium borohydride were added in portions to a solution of 209 mg (0.5 mmol) of Compound No. 70 in 30 mt of ethanol. The reaction mixture was stirred for one hour at 0 °C and for further 16 hours at room temperature.

The reaction mixture was post-treated as in Example 23. The residue was then purified by chromatography on a silica gel column (eluent: ethyl acetate), whereby 201 mg of the title compound were obtained (yield: 96%).

Example 78

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Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 78)

A suspension of 69 mg (0.27 mmol) of Compound No. 26, 58 mg (0.32 mmol) of 1-(4-fluorophenyl)piperazine, 45 mg (0.54 mmol) of sodium bicarbonate and 81 mg (0.54 mmol) of sodium iodide in 5 ml of
acetonitrile was refluxed for 20 hours. The reaction mixture was post-treated and purified as in Example 63,
whereby 97 mg of the title compound were obtained (yield: 89%).

Example 79

Synthesis of 2-[3-[4-(4-fluorophenyl)piperazin-1-yl]propyl]-3,4-dihydro-2H-1,2-benzothiazine 1,1-dioxide (Compound No. 79)

As in Example 78, the reaction, post-treatment and purification were conducted employing 73 mg (0.28 mmol) of Compound No. 27, 60 mg (0.34 mmol) of 1-(4-fluorophenyl)piperazine, 47 mg (0.56 mmol) of sodium bicarbonate, 84 mg (0.56 mmol) of sodium iodide and 5 m1 of acetonitrile, whereby 108 mg of the title compound were obtained (yield: 96%).

The structural formulas and physical properties of the compounds obtained in the above examples are summarized in the following Table 1.

5	IR (cm ⁻¹) (): messuring method	(KBr.) 3253, 2965, 2887, 1427, 1322, 1175, 1144, 1076, 1026, 953, 815, 764	(KBr.) 3245, 2977, 1426 1382, 1314, 1256, 1158, 1077, 944, 892, 820, 698	(KBr.) 3295, 1405, 1318. 1167, 1137, 1052. 1015, 824, 762. 702
10		(KBr.) 3253, 1427, 1144, 953,	(KBr.) 3245, 1382, 1158, 892,	
15	NHR (6 ppm)* (): observation frequency	(2704Hz) 1.58(1H,d.quint,J=2.3Hz,13.9Hz), 2.27(1H,m), 3.96-4.20(6H,m), 5.07 (1H,t,J=7.6Hz), 7.49(1H,dt,J=1.3Hz,7.9Hz), 7.59(1H,dt,J=1.3Hz,7.9Hz), 7.746(1H,dd,J=1.3Hz,7.9Hz), 7.87(1H,dd,J=1.3Hz,7.9Hz),	(270MHz) 1.33(3H,d,J-5.9Hz), 1.40(3H,d,J-5.9Hz), 3.74(2H,dd,J-1.3Hz,7.9Hz), 3.79-3.97(2H,m), 5.22(1H,t,J-7.9Hz), 7.48-7.52(3H,m), 7.77(1H,d,J-7.9Hz)	(270Mhz) 3.29(1H.m), 3.41(1H.m), 3.92(2H.m), 4.21(1H.m), 4.52(1H.m), 5.22(1H.t, J-7.3Hz), 7.49(1H.m), 7.58(1H.m), 7.77(1H.m), 7.80(1H.m)
20		(270MHz) 1.58(1H,d.c 2.27(1H,m) (1H,t,J=7.(1H,m) 1.3Hz,7.9H 7.9H2), 7.3	(270MHz) 1.33(3H,d., J=5.9Hz), 7.9Hz), 3. (1H,t,J=7.	(270HHz) 3.29(1H,m) 4.21(1H,m) J=7.3Hz), 7.77(1H,m)
Table 1	Property Melting point (recrystalli- zation solvent)	Coloriess prism crystals 147.0-148.5°C (ethyl acetate -hexane)	Colorless prism crystals 142.5-143.0°C (ethyl acetate -hexane)	Coloriess prism crystals 169.5-171.5°C (ethyl acetate -hexane)
35	Pr He) (re	Col 147 147	[] [] [] [] [] [] [] [] [] []	O 15 8 9 4
35	ructural formula	Coverage in o	CONT. NO.	Cover in a
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4 5	comp'd No.	-	~	m
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* Measured in CDCl₃ with IMS as an internal standard unless otherwise specifically indicated.

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		IR (cm ⁻¹) (): measuring method	1318, 853, 668	1452. 1390 1176 1058. 852 754.	1325. 840. 667
5		IR (cm ⁻¹) measuring	(KBr.) 3263, 1402, 1318, 1164, 1131, 853, 751, 708, 668	1471. 1418. 1276. 1085. 890. 779.	(KBr.) 3281, 1405, 1325, 1170, 1068, 840, 753, 711, 667
		II ()	(KBr.) 3263, 1164, 751,	(KBr.) 3274. 1428. 1321. 1133. 998. 818. 709.	(KBr.) 3281. 1170. 753.
10					
		quency	(2н. d.	t ,	5.49 =1.3Hz. :9Hz).
15		n)* on frec), 3.96(27.9kz), 1kz), 1kz), 1kz), 3kz),), 2.78 (2H,d,J) (2H,d,J) (2H,d,J) (3Hz), (3Hz)	.40-2.6 .9Hz). IH,dt,J I.3Hz,7
		NHR (6 ppm)* (): observation frequency	14(2H,m) 1H, t, Ja 3Hz, 7.5 3Hz, 7.5 3Hz, 7.5	3(1H,m) 1, 4.24 4,t,3=7 4Hz,7.1 3Hz,7.1 9Hz,7.1	6Hz), 2 H,d,J=7 , 7.46(H,dt,J= .3Hz,7.
	-	NMR (): ob	(270HHz) 3.45(2H,m), 3.64(2H,m), 3.96(2H,d, 3-7.9Hz), 5.39(1H,t,3-7.9Hz), 7.42(1H,dt,3-1.3Hz,7.9Hz), 7.54(1H,dt,3-1.3Hz,7.9Hz), 7.74(1H,dd,3-1.3Hz,7.9Hz), 8.04(1H,dd,3-1.3Hz,7.9Hz)	(400Mkz) 2.01(1H.m), 2.23(1H.m), 2.78(2H, dt. J-3.7hz,14.7hz), 3.11(2H, dt. J-2.5hz,14.7hz), 4.24(2H, d. J-2.7hz), 5.29(1H, t. J-7.7hz), 7.47(1H, dt. J-1.4hz,7.8hz), 7.57(1H, dt. J-1.4hz,7.8hz), 7.87(1H, dd. J-1.3hz,7.8hz), 7.87(1H, dd. J-1.3hz,7.8hz), 8.29(1H, dd. J-0.9hz,7.8hz), 8.20(1H, dd. J-0.9hz,7.8hz),	(2704Hz) 1.18(6H,t,J=7.6Hz), 2.40-2.67 (4H,m), 4.01(2H,d,J=7.3Hz), 5.49 (1H,t,J=7.9Hz), 7.46(1H,dt,J=1.3Hz,7.9Hz), 7.9Hz), 7.58(1H,dt,J=1.3Hz,7.9Hz), 7.82(1H,dd,J=1.3Hz,7.9Hz), 8.03 (1H,dd,J=1.3Hz,7.9Hz)
20	nt'd		(270HHz) 3.45(2H, 3-7.9Hz) 7.42(1H, 7.54(1H, 7.74(1H,	(400MKZ) 2.01(1H. dt.J=3.7 J=2.5KZ, 7.7KZ), 7.47(1H, 7.57(1H, 8.09(1H,	(270Mkz) 1.18(6H, (4H.m). (1H, t.)- 7.9Hz). 7.82(1H, (1H, dd,
	Table 1 (Cont'd)		33 7 7 8 8	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
25	ble	nt 11- rent)	awdery	orism ;°C ie)	needle C tate
	Ta	Property Melting point (recrystalli- zation solvent)	Colorless powdery crystals 231.5-232.0°C (chloroform -hexane)	Colorless prism crystals 231.0-232.5°C (acetonitrile)	Colorless needle crystals 86.5-87.0°C (ethyl acetate -hexane)
30		Pro Melt (rec zati	Colo crys 231. (chl	Colo crys 231. (ace	Cold Crys 86.9 (ett
35					
		= -	_	_	ž, Ž
40		Structural formula	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	~ ~ ±	X %o
		St.			c, H, S
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 ullet Measured in CDCl $_{
m J}$ with TMS as an internal standard unless otherwise specifically indicated.

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		IR (cm ⁻¹) (): measuring method	1446, 1152, 969, 777,	KBr) 3300-3100, 1630, 1560, 1400, 1310, 1250, 1150, 1070, 990, 870, 760, 670	(KBr.) 3258, 2823, 1410. 1316, 1173, 1064, 884, 810, 709
5		IR (cm ⁻¹) measuring	2976, 1225, 1040, 821,	100. 1400. 1150. 870.	2623, 1173, 810,
	-	():	(KBr.) 3256, 2976, 1446, 1328, 1225, 1152, 1085, 1040, 969, 898, 821, 777, 693	(KBr.) 3300-3100, 1560, 1400, 1250, 1150, 990, 870, 670	(KBr.) 3258, 1316, 884,
10					
15		NMR (6 ppm)* (): observation frequency	(270Mtz) 3.25(6H,s), 3.82(2H,d.J=7.9Hz), 5.11(1H,br.d), 7.51-7.60(2H,m), 7.72(1H,m), 7.83(1H,m)	(2704Hz) 3.79(3H,s), 5.96(1H,br.s), 6.03 (1H,d,J=2.6Hz), 7.54(1H,dt,J=1.3Hz, 7.3Hz), 7.64(1H,dt,J=1.3Hz,7.9Hz), 7.83(1H,m)	(270MHz) 3.43(3H,s), 3.78(1H,m), 4.01(1H,m), 4.10(1H,m), 5.05(1H,br), 7.38(1H,dd, J-2.0Hz,7.3Hz), 7.53-7.56(2H,m), 7.87(1H,dd,J-2.0Hz,7.3Hz)
20	Table 1 (Cont'd)	¥ ()	(270Mhz) 3.25(6H,s), 3.82(2H,d., 5.11(1H,br.d), 7.51-7.0	(270HHz) 3.79(3H,s), 5 (1H,d,J=2.6Hz 7.3Hz), 7.64(7.83(1H,m)	(270HHz) 3.43(3H.s), 3 4.10(1H.m), 5 J-2.0Hz,7.3H
25					ا و
30	Table	Property Helting point [recrystalli- zation solvent]	Colorless prism crystals 161.0-162.0°C (ethyl acetate -hexane)	Colorless prism crystals 85.0-86.0 °C (ethyl acetate -hexane)	Colorless needle crystals 80.0-81.0°C (ethyl acetate -hexane)
35					
40	·	Structural formula	O.H.O.	HO HE SO	ž vo
45		ļ			
50		Comp'd No.	,	∞	on

* Measured in COCl₃ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm^{-1}) []: measuring method	(KBF) 1440, 1327, 1262, 1154, 1091, 1051, 1031, 994, 945, 895, 847, 771, 754, 687	(KBr.) 2957, 1460, 1330, 1228, 1174, 1149, 1051, 995, 950, 937, 799, 776	(film) 1331, 1155, 1052, 948, 764, 691
15		NMR (6 ppm)* (): observation frequency), 3.98(2H.s),), 4.23-4.38 67(3H.m),	(270MHz) 2.11(2H,m), 3.62(2H,t,J=6.6Hz), 3.72(2H,t,J=5.9Hz), 3.91(2H,s), 4.17(2H,m), 4.29(2H,m), 7.46- 7.64(3H,m), 7.79(1H,m)	(2704Hz) 1.69-2.00(4H,m), 3.47(2H,t,J= 5.9Hz), 3.60(2H,t,J=5.9Hz), 3.87(2H,s), 4.07-4.24(2H,m), 4.24-4.38(2H,m), 7.40-7.67 (3H,m), 7.78(1H,m)
20	Table 1 (Cont'd)	NM (): ol	(270Hhz) 3.60-3.86(4H,m), 3.98(2H,s), 4.09-4.23(2H,m), 4.23-4.38 (2H,m), 7.41-7.67(3H,m), 7.77(1H,d,J=7.9Hz)	(2704Hz) 2.11(2H,m), 3.62(2H,t., 3.72(2H,t.,J=5.9Hz), 3.4 4.17(2H,m), 4.29(2H,m) 7.64(3H,m), 7.79(1H,m)	(270HHz) 1.69-2.00(4H,m), 3.47(2H,t, 5.9Hz), 3.60(2H,t,J=5.9Hz), 3.87(2H,s), 4.07-4.24(2H,m) 4.24-4.38(2H,m), 7.40-7.67 (3H,m), 7.78(1H,m)
25 30	Table 1	Property Melting point (recrystalli- zation solvent)	Colorless prism crystals 126.5-128.0°C (ethyl acetate -hexane)	Colorless prism crystals 72.5-74.0°C (ethyl acetate -hexane)	Coloriess oil
35					מים
40		Structural formula	O (CH ₁) ₁ Cl	0 N	0 × 20 × 20 × 20 × 20 × 20 × 20 × 20 ×
45		P			21
50		Comp'd No.	10	=	_

 ullet Measured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

		ie thod	y	430, 153, 766,	1474. 1319. 1133. 912. 758. 678
5		IR (cm ⁻¹) measuring n	1319, 1163, 942, <i>77</i> 6 652	1471, 1430, 1296, 1153, 790, 768, 654	(KBr.) 3530, 2907, 1474, 1442, 1425, 1319, 1228, 1165, 1133, 1094, 952, 912, 885, 784, 758, 720, 696, 678
	,	IR (cm ⁻¹) (): measuring method	(KBr.) 2931. 13 1089. 9 757. 6	(KBr.) 2920, 1471, 1430, 1350, 1296, 1153, 953, 790, 768, 746, 654	(KBr.) 3630. 2 1442. 1 1228. 1 1094, 885, 720.
10					
15		. NMR (δ ppm)* (): observation frequency	(400MHz) 3.49(2H,m), 3.76(2H, t,J=6.6Mz), 3.92(2H,t,J=6.6Mz), 4.21(2H,s), 7.40(1H,m), 7.52(1H, m), 7.75(1H.dd,J=1.2Mz,7.9Mz), 7.95(1H,d,J=8.2Hz)	(2004Hz) 2.17(2H.quint.J=6.3Hz), 3.48 (2H.m), 3.64(2H.m), 3.70(2H.t, J=6.3Hz), 3.75(2H.t,J=6.3Hz), 4.10(2H.s), 7.40(1H.dt,J=1.2Hz, 7.9Hz), 7.52(1H.dt,J=1.2Hz, 7.9Hz), 7.56(1H,dd,J=1.2Hz,7.9Hz), 7.56(1H,dd,J=1.2Hz,7.9Hz)	(400HHz) 2.03(1H,m), 2.26(1H,m), 2.83(2H,dt,J=3.8Hz,14.8Hz), 3.15(2H,ddd,J=2.5Hz,12.7Hz,14.8Hz), 3.75(2H,t,J=6.2Hz), 4.62(2H,s), 7.46(1H,dt,J=1.3Hz,7.8Hz), 7.82(1H,dd,J=1.3Hz,7.8Hz), 7.82(1H,dd,J=1.3Hz,7.8Hz), 7.82(1H,dd,J=1.3Hz,7.8Hz), 8.09(1H,dd,J=1.3Hz,7.8Hz),
20	Table 1 (Cont'd)	. :	(400Mt2) 3.49(2H,m), 3.64(2l, 1,1=6.6Hz), 3.92(2,4.2l(2H,s), 7.40(1l, m), 7.75(1H,dd,1=1.7.95(1H,d,1=8.2Hz))	(400MHz) 2.17(2H.quin (2H.m), 3.64 J-6.3Hz), 3.64 4.10(2H.s), Hz,7.9Hz), 7.76 Hz), 7.95(1)	(400MHz) 2.03(1H.m), dt,J=3.8Hz, J=2.5Hz,12. t,J=6.2Hz), 4.62(2H,s), 7.8Hz), 7.82(1H,dd,
25	le 1	1	Es	es ر بور	te c sm
30	Tab]	Property Melting point (recrystalli- zation solvent)	Colorless prism crystals 120.5-121.0°C (ethyl acetate -hexane)	Colorless prism crystals 137.0-137.5°C (ethyl acetate -hexane)	Coloriess prism crystals 198.0-200.0°C (ethyl acetate -hexane)
35					
40		Structural formula	S S S S S S S S S S S S S S S S S S S	S S S S S S S S S S S S S S S S S S S	S S N S S S S S S S S S S S S S S S S S
45					
50		Comp'd No.	E	3	51

 st Heasured in CDCl $_{
m J}$ with IMS as an internal standard unless otherwise specifically indicated.

5	IR (cm ⁻¹) (): Reasuring method	(KBr) 2902, 1438, 1424 1300, 1241, 1163, 1136, 1076, 962, 884, 778, 754 722, 700, 674, 656	(film) 2947, 1459, 1337, 1142, 1064, 986, 936, 768, 688	(film) 1630, 1440, 1380, 1340, 1270, 1230, 1170, 1130, 1070, 770
10		2.23		
15	NHR (6 ppm) [*] (): observation frequency	(4000Hz) 2.04(1H,m), 2.14(2H,quint,J=6.2Hz), 2.23 (1H,m), 2.82(2H,dt,J=3.7Hz,14.7Hz), 3.17(2H,ddd,J=2.4Hz,12.7Hz,14.7Hz), 3.75(2H,t,J=6.2Hz), 3.79(2H,t,J=6.2 Hz), 4.49(2H,s), 7.45(1H,dt,J= 1.0Hz,7.8Hz), 7.55(1H,dt,J=1.3Hz, 7.8Hz), 7.85(1H,dd,J=1.3Hz, 8.07(1H,dd,J=1.0Hz,7.8Hz))	(270MHz) 2.21(2H, H, 3.27(6H, s), 3.58 (2H, t, J=6.6Hz), 3.63(2H, t, J= 6.6Hz), 3.99(2H, s), 7.49-7.60 (2H, M), 7.69(1H, dd, J=2.0Hz, 7.3Hz), 7.85(1H, dd, J=2.0Hz,	(270MHz) 2.17(2H,m), 3.39(2H,t,J=6.6Hz), 3.77(2H,t,J=6.6Hz), 3.82(3H,s), 6.06(1H,s), 7.55(1H,dt,J=1.3Hz, 7.9Hz), 7.64(1H,dt,J=1.3Hz, 7.9Hz), 7.81(1H,dd,J=1.3Hz, 7.9Hz), 7.85(1H,dd,J=1.3Hz,
% % Table 1 (Cont'd)	:()	(400HLZ) 2.04(1H,m), 2.82 (1H,m), 2.82 3.17(2H,ddd, 3.75(2H,t.)=HZ), 4.49(2H,t.) 1.0HZ,7.8HZ) 7.8HZ), 7.85	(270HHz) 2.21(2H m), (2H.t.J-6.6H 6.6Hz), 3.99 (2H,m), 7.65 7.3Hz), 7.88	(270kHz) 2.17(2H·m). 3.77(2H·t, J·6.06(1H·s). 7.9Hz), 7.6i. 7.9Hz), 7.8i. 7.9Hz), 7.8i.
Table 1	Property Kelting point (recrystalli- zation solvent)	Colorless prism crystals 161.0-163.0°C (ethyl acetate -hexane)	Colorless oil	Colorless oil
30	Pro Relt (rec	Colo crys 161. (eth	00 .	3
35	ıral ila	າດແ _ປ ງຕ	осн, , к. к. с. н. д. вг	, (CH ₃₎ ,8r
40	Structural	w Sing	g vo	8 - , , , , ,
45				
	Comp'd No.	91	=	81
50	i	ł .	1	•

 * Heasured in COCl $_{
m J}$ with TMS as an internal standard unless otherwise specifically indicated.

•				
5	IR (cm ⁻¹) (): measuring method	(film) 1460, 1450, 1340, 1170, 1130, 1070, 760, 700	(film) 2937, 2828, 1445, 1336, 1170, 1133, 1071, 946, 761, 703	(KBr.) 1698, 1588, 1341, 1281, 1235, 1175, 1120, 928, 759
10			_	
15	NMR (& ppm)* (): observation frequency	(270MHz) 2.19(2H,m), 3.43-3.62(4H,m), 3.56(3H,s), 3.88(1H,dd,J=4.0Hz, 15.2Hz), 4.15(1H,dd,J=4.0Hz, 15.2Hz), 4.35(1H,t,J=4.0Hz), 7.46-7.61(3H,m), 7.83(1H,dd,J= 1.3Hz,7.9Hz)	(270Hhz) 2.11(2H,m), 3.51(2H,m), 3.59(3H,s), 3.71(2H,m), 3.87(1H,dd,J=4.0Hz,15.2Hz), 4.14(1H,dd,J=4.0Hz,15.2Hz), 4.35(1H,t,J=4.0Hz), 7.45-7.60 (3H,m), 7.83(1H,dd,J=1.3Hz, 7.9Hz)	(270Mhz) 2.08(2H.quint, J=6.6Hz), 3.36(2H.t., J=6.6Hz), 4.46 (2H.s), 7.75(1H.ddd, J=1.3Hz, 7.3Hz, 7.6Hz), 7.89(1H.ddd, J=1.3Hz, 7.3Hz, 7.6Hz), 7.89(1H.dd, J=1.3Hz, 7.6Hz), 8.08(1H.dd, J=1.3Hz, 7.6Hz),
20 g	_	(270MHz) (1.19(2H.m), (3.56(3H.s), (5.2Hz), 4.1 (5.2Hz), 4.3 (7.46-7.51(3Hz), 1.3Hz,7.9Hz)	(270HHz) 2.11(2H.m), 3.71(2H.m), 3.87(1H.dd., 4.14(1H.dd., 4.35(1H.t.J., (3H.m), 7.81	fftz) (2H.q. (2H.q. 5Hz), z), 7, z), 7, z), 7 z), 7 z), 7 z), 7 z), 7
% & & Cat'd)		(270MK2) 2.19(2H, 3.56(3H, 15.2K2), 15.2K2), 7.46-7.6	(270Mt2) 2.11(2H, 3.71(2H, 3.87(1H, 4.14(1H, 4.35(1H, 3.35(1H, 3.3	(270Mhz) 2.08(2H,) J=6.6Hz) (2H, s), 7.6Hz), 7.6Hz), 8.08(1H,
25 0	nt 1- ent)	=	=	rrism C cate
	Property Helting point (recrystalli- zation solvent)	Colorless oil	Colorless of 1	Colorless prism crystals 76.5-78.0°C (ethyl acetate -hexane)
30				
35	Structural	0.04, 0.00 (CH _{3), 81}	ock, 20, 20, 20, 20, 20, 20, 20, 20, 20, 20	0 = \(\frac{1}{2} \cdot \frac
40	Str			
45				
	Comp d	<u>s</u>	90	2
50	I	l	1	ı

 ullet Heasured in CDCl $_{
m 3}$ with TMS as an internal standard unless otherwise specifically indicated.

5	IR (cm ⁻¹) (): measuring method	(film) 3422, 1341, 1173, 1131, 955, 759	(film) 3484, 1444, 1323, 1168, 1132, 1059	(film) 2975, 2875, 1446, 1337, 1172, 1071, 762, 706
10				
15	KMR (6 ppm) (): observation frequency	(270MHz) 2.07(2H,quint,J=6.6Hz), 3.17 (2H,t,J=6.6Hz), 3.65(2H,t,J=6.6Hz), 4.71(2H,s), 7.51-7.69 (2H,m), 7.87(1H,m), 7.99(1H,m), 8.77(1H,s)	(270HHz) 2.11(2H,quint,J-6.3Hz), 2.83 (1H,d,J=7.6Hz), 3.37-3.65(2H,m), 3.67(2H,t,J-6.3Hz), 3.74(1H,dd, J-5.3Hz,14.5Hz), 4.02(1H,dd,J-6.6 Hz,14.5Hz), 4.85(1H,m), 7.42-7.68 (3H,m), 7.90(1H,d,J=7.9Hz)	(270MHz) 1.30(3H,t,J=7.3Hz), 2.11(2H,m), 3.51(2H,m), 3.60-3.90(5H,m), 4.14 (1H,dd,J=4.0Hz,15.2Hz), 4.46 (1H,t,J=4.0Hz), 7.40-7.60(3H,m), 7.83(1H,dd,J=1.3Hz,7.9Hz)
20 T	_	(270MHz) 2.07(2H,qu (2H,t,J=6. 6.6Hz), 4. (2H,m), 7. (2H,m), 7.	(2704Hz) 2.11(2H.qu (1H,d.)=7. 3.67(2H,t J=5.3Hz,1. Hz,14.5Hz (3H,m), 7	(270MHz) 1.30(3H,t 3.51(2H,m (1H,dd,J= (1H,t,J=4) 7.83(1H,d
Cont		(270 2.07 (2H. (2H. (2H.)	(27c 2.11 2.11 (1H, 3.67 3.67 J=5. Hz,1	(27/ 1.3(3.51 3.51 (114, (114,
1 (6				
% & & & & & & & & & & & & & & & & & & &	Property Melting point (recrystalli- zation solvent)	Yel low oil	Colorless of 1	Coloriess of 1
30	Pro Helt (rec zati	rel]	90	Colo
35	·	-	_	5
	_	cห _ม เต	o first	(CH _J),C
40	Structural formula	NOH N. S. N. S.	# \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	0C,H1 S. "(CH3)A
45				
	Comp'd No.	22	23	\$2
50				Į į

 ullet Heasured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method	(†111) 1455, 1335, 1169 1133, 1067, 760, 702	(f11m) 1621, 1477, 1442, 1332, 1174, 1133, 1069, 778, 752	(film) 2963, 1479, 1446, 1317, 1161, 1069, 948, 856, 760, 710, 678
10					
15		NMR (6 ppm)* (): observation frequency	(270HHz) 2.08(2H,quint,J=5.3Hz), 3.52 (2H,m), 3.69(2H,m), 3.88(1H, dd,J=4.0Hz,15.2Hz), 4.13(1H,dd, J=4.0Hz,15.2Hz), 4.57(1H,t,J= 4.0Hz), 4.70(1H,d,J=11.9Hz), 4.79(1H,d,J=11.9Hz), 7.30-7.60 (8H,m), 7.83(1H,dd,J=2.0Hz,7.9Hz)	(400MHz) 2.22(2Hm), 3.61(2H,t,J=6.1Hz), 3.93(2H,t,J=6.5Hz), 6.24(1H,d, J=7.9Hz), 6.63(1H,d,J=7.9Hz), 7.37(1H,d,J=7.8Hz), 7.47(1H,m), 7.58(1H,m), 7.91(1H,d,J=7.9Hz)	(400HHz) 2.13(2H,t,J=6.3Hz), 3.35(2H,t,J=6.6Hz), 3.70(2H,t, J=6.1Hz), 3.92(2H,t,J=6.3Hz), 7.23(1H,d,J=7.7Hz), 7.37(1H,m), 7.46(1H,m), 7.83(1H,d,J=7.8Hz)
20	Table 1 (Cont'd))	(2704Hz) 2.08(2H,qu, (2H,m), 3.dd, 3=4.0Hz, 1! 4.0Hz, 1! 4.0Hz, 1! 4.0Hz, 1! (8H,m), 7	(400Hhz) 2.22(2H.m. 3.93(2H.t. 3.97(1H.d. 7.37(1H.d.	(400Hz) 2.13(2H,π 3.35(2H,t) 3.46(1H,t) 7.23(1H,t) 7.46(1H,t)
25	H				
30	Table	Property Helting point (recrystalli- zation solvent)	Colorless oil	Coloriess oil	Colorless oil
35					
40		Structural	10°(4'H), 10°	S. '(CH1),CI	S. 'CHJ).
50		Comp'd No.	\$2	98	ız

 ullet Heasured in CDC $_{
m J}$ with IMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method	(KBr) 1597, 1499, 1326, 1158, 1052, 952 764, 755	(KBr) 2824, 1509, 1324, 1236, 1154, 1051, 947, 825, 755	(KBr) 1508, 1329, 1210. 1151, 1052, 830. 763
		<u>ہ</u>		1 - 6	
15	1)	NMR (6 ppm)* (): observation frequency	(270MHz) 1.86(2H,m), 2.51(2H,t,J=6.6Hz), 2.61(4H,m), 3.20(4H,m), 3.53(2H, t,J=6.6Hz), 3.92(2H,s), 4.07-4.21 (2H,m), 4.21-4.35(2H,m), 6.85(1H, t,J=7.3Hz), 6.93(2H,d,J=8.6Hz), 7.26(2H,dd,J=7.3Hz,8.6Hz), 7.65(3H,m), 7.79(1H,m)	(2704Hz) 2.50-2.80(6H.m), 3.10(4H.m), 3.51 (2H.t.J=6.6Hz), 4.02(2H.s), 4.07- 4.23(2H.m), 4.23-4.38(2H.m), 6.77- 7.04(4H.m), 7.40-7.65(3H.m),	(270HHz) 1.50-1.78(4H.m), 2.43(2H.t.)= 7.0Hz), 2.59(4H.m), 3.11(4H.m), 3.46(2H.t.)=7.0Hz), 3.88(2H.s), 4.05-4.22(2H.m), 4.22-4.38(2H.m), 6.77-7.05(4H.m), 7.42-7.05(3H.m), 7.79(1H.m)
	nt 'd		(270MHz) 1.86(2H, 2.61(4H, 1.5-6.6H (2H,m), 1.3-7.3H 7.26(2H,	(270HHz) 2.50-2.80((2H.t.J=6. 4.23(2H.m) 7.04(4H.m)	(270Hkz) 1.50-1.78(7.0Hz), 2. 3.46(2h,t, 4.05-4.22(6.77-7.05(7.79(1H,m)
	Ö)		3161616	22247	
25	Table 1 (Cont'd)	Property Melting point (recrystalli- zation solvent)	Colorless needle crystals 113.0-115.0°C (2-propanol -isopropyl ether)	Colorless prism crystals 130.0-131.5°C (ethyl acetate -hexane)	Pale brown prism crystals 120.5-121.5°C (ethyl acetate -hexane)
30		Proj Helt (reci	Colorless crystals 113.0-11: (2-propar	Colorless crystals 130.0-131 (ethyl achexane)	Pale pris 120. (eth
35					
40	·	Structural	S. 'CCH _{2), N} (CH _{2), N}	CHJIN N	S. V. (CH.), N.
45		!			
50		Comp 'd No.	28	53	30

 st Measured in CDCl $_{3}$ with IMS as an internal standard unless otherwise specifically indicated.

		ae thod	1156. 135.	155.	. 574, .332, .053.
5		IR (cm ⁻¹) (): measuring method	1333, 1 1032, 768, 692	(film) 2836, 1332, 1155, 1052, 948, 762, 700	(KBr.) 1897, 1602, 1574, 1496, 1458, 1332, 1202, 1157, 1053, 768, 699
		Ë	(KBr.) 2813. 1054. 952. 715.	(film) 2936, 1052, 700	(KBr.) 2947, 1496, 1202, 768,
10	1	luency	7.46-	ون (ع. (ع. ۲۵), (ع. ۲۳),	(2H, t, H,m). s). (2H,m). Je 9Hz).
15		NHR (6 ppm)* (): observation frequency	(270ML) 1.80(2H.m), 2.28-2.62(10H.m), 3.47(2H.m, J=6.6K2), 3.88(2H.s), 4.10(2H.m), 4.23(3H.m), 7.13- 7.34(6H.m), 7.36-7.46(4H.m), 7.46- 7.62(3H.m), 7.77(1H.d.J=7.3Hz)	(270Mtz) 1.66-1.96(6H.m), 2.07(2H.dt.Ja- 2.8Hz, 11.5Hz), 2.40-2.59(3H.m), 3.05(2H.m), 3.51(2H.t.Ja-6.9Hz), 3.92(2H.s), 4.08-4.21(2H.m), 4.21-4.35(2H.m), 7.14-7.37(5H.m), 7.44-7.64(3H.m), 7.78(1H.m)	(270whz) 1-86(2H,quint, J=7.3Hz), 2.51(2H,t, J=7.3Hz), 2.50(4H,m), 3.53(2H,t), 3.53(2H,t), 3.39(3H,s), 3.53(2H,s), 4.16(2H,m), 4.27(2H,m), 6.39-6.49(2H,m), 6.53(1H,dd,J=1.3Hz,7.9Hz), 7.17(1H,t,J=7.9Hz), 7.47-7.62(3H,m), 7.80(1H,d,J=7.6Hz)
20	Table 1 (Cont'd)	•	(270MKz) 1.80(2H,n 3.47(2H,t 4.10(2H,n 7.34(6H,n	(270HHz) 1.66-1.96 2.8Hz, 11 3.05(2H,n 4.21-4.31 7.44-7.6	(270HNz) 1.86(2H, 4) 1.86(2H, 4) 1.312(2H, 3) 3.92(2H, 6) 6.39-6.4 1.3Hz,?.
25	Table 1	Property Melting point (recrystalli- zation solvent)	Colorless powdery crystals 93.5-95.5°C (ethanol)	Pale yellow oil	Colorless powdery crystals 109.0-110.0°C (ethyl acetate -hexane)
30			•		4
35		Structural formula		0, CCH4J/N	
40				X vo.	
4 5		Comp d	i i	32	s

 ullet Measured in CDCl $_{
m J}$ with TMS as an internal standard unless otherwise specifically indicated.

48

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5	IR (cm ⁻¹) (): measuring method	(KBr.) 1508, 1337, 1224, 1161, 1054, 826, 767, 694	(KB-) 2819, 1501, 1445, 1330, 1231, 1162, 1052, 948, 769, 752, 693	(KBr) 1608, 1496, 1323, 1255, 1154, 1124, 1049, 996, 973, 760, 691
10	requency		52(2H, 12 30(2H,m), :(3H,m),	50(2H, 20 1, 3.92 (H,m).
15	NHR (6 ppm)* (): observation frequency	(270MHz) 1.86(2H,quint,J=6.9Hz). 2.51(2H,t,J=7.3Hz), 2.62 (4H,m), 3.10(4H,m), 3.52 (2H,t,J=6.9Hz), 3.77(3H,s). 3.92(2H,s), 4.17(2H,m), 4.27(2H,m), 6.80-6.95(4H,m), 7.47-7.62(3H,m), 7.79(IH,m)	(2704Hz) 1.86[2H.quint.J=7.3Hz], 2.52[2H, t.J=7.3Hz), 2.64[4H.m], 3.12 [4H.m], 3.53[2H.t.J=7.3Hz], 3.92[2H.s], 4.18[2H.m], 4.30[2H.m], 6.88-7.10[4H.m], 7.46-7.52[3H.m], 7.80[1H.m]	(270MHz) 1.86(2H,quint,J=6.9Hz), 2.50(2H, 1.J=6.9Hz), 2.60(4H,m), 3.20 (4H,m), 3.52(2H,t,J=6.9Hz), 3.92 (2H,s), 4.15(2H,m), 4.29(2H,m), 6.47-6.70(3H,m), 7.18(1H,m), 7.47- 7.62(3H,m), 7.79(1H,m)
% & & & & & & & & & & & & & & & & & & &)	(270Mhz) 1.86(2H,qu 2.51(2H,t, (4H,m), 3, (2H,t,J=6, 3.92(2H,s) 7.47-7.62((270Mhz) 1.86(2H.qu t,J=7.34z) (4H,m), 3. 3.92(2H,s) 6.88=7.10(7.80(1H,m)	(270Mt) 1.86(2H.q t,J=6.9hz (4H.m), 3 (2H.s), 4 (2H.s), 6.47-6.70
% Table 1	Property Melting point (recrystalli- zation solvent)	Pale yellow needle crystals 119.0-119.5°C (ethyl acetate -hexane)	Colorless prism crystals 70.0-71.5°C (ethyl acetate -hexane -ethyl ether)	Colorless plate crystals 154.0-155.0°C (ethyl acetate -hexane)
30	Prop Melti (reci		Colorless Crystals 70.0-71. (ethyl ac -hexane -ethyl el	Colo Crys 154, (eth
35	å å	, N — OCH3	Ç	
40	Structural	S. X. (CH ₂), N. X. Y. C. V. S. Y.	S. Y.	N. Liebay.
45	7			
50 .	Comp'd	***	35	36

Heasured in CDCl₃ with TMS as an internal standard unless otherwise specifically indicated.

5	IR (cm ⁻¹) (): measuring method	(KBr.) 3676, 2820, 1514, 1457, 1332, 1230, 1164, 1052, 964, 824, 763, 692	(KBr.) 1498, 1331, 1305, 1232, 1175, 1154, 1050, 938, 818, 772, 734, 696	(film) 3264, 2949, 2823, 1609, 1514, 1453, 1388, 1329, 1237, 1154, 1052, 974, 824, 766, 736, 692
10				
75 (p	NMR (6 ppm)* (): observation frequency	(270NHz) 1.86(2H.m), 2.51(2H.t.J=7.3Hz), 2.62(4H.m), 3.09(4H.m), 3.52(2H, t.J=6.9Hz), 3.92(2H.s), 4.16 [2H,m), 4.27(2H.m), 6.76(2H.m), 6.84(2H,m), 7.48-7.62(3H,m), 7.79(1H,m)	(270MHz) 1.86(2H.quint.J=7.0Hz), 2.50 (2H,t,J=7.0Hz), 2.50(4H.m), 3.16 (4H.m), 3.52(2H.t,J=7.0Hz), 3.92 (2H.s), 4.16(2H.m), 4.29(2H.m), 6.83(2H,m), 7.20(2H.m), 7.46- 7.62(3H,m), 7.80(1H.m)	(400HHz) 1.86(ZH,quint,J=7.0Hz), 2.51 (ZH,t,J=7.0Hz), 2.61(4H,m), 2.94(3H,s), 3.19(4H,m), 3.53(ZH,t,J=7.0Hz), 3.91(ZH,s), 4.15(ZH,m), 4.29(ZH,m), 6.11(1H,br.s), 6.89(ZH,d,J=9.0Hz), 7.15(ZH,d,J=9.0Hz), 7.15(ZH,d,J=9.0Hz), 7.48-7.60(3H,m), 7.79 (1H,d,J=7.9Hz)
ont '		(270MHz) 1.86(2H.m) 2.62(4H.m) t.,1=6.9Hz) (2H.m). 4. (2H.m). 4.	(270MHz) 1.86(2H, (2H,t.)= (4H.m), (2H.s), (2H,s), 6.83(2H,	(400HHz) 1.86(2H, (2H, L, J= 2.94(3H, t, J=7.0H m), 4.29 5.89(2H, J=9.0Hz) (1H, d, J=
၁၁)				
25 H	ਹੁ	8 J	و ر الا	
S S S S S S S S S S S S S S S S S S S	Property Helting point (recrystalli- zation solvent)	Colorless prism crystals 184.5-185.5°C (methanol -ethyl ether)	Colorless prism crystals 118.5-119.5°C (ethyl acetate -hexane)	Pale red oil
35		¥ 0	Ū Z	S. NHSO,CH,
40	Structural	(ch.b) (ch.b) (c)	, C. C. L.	(Purchas)
45				
50	Comp'd No.	33	88	85

 * Measured in CDCl $_{
m 3}$ with TMS as an internal standard unless otherwise specifically indicated.

	•			
5	IR (cn^{-1}) (): measuring method	(KBr) 1680, 1595, 1325, 1211, 1183, 1158, 1055, 953, 758	(film) 2944, 2804, 1682. 1611, 1494, 1446. 1334, 1152, 1054, 946, 758, 735. 694	(film) 2944, 2812, 1614, 1337, 1272, 1243, 1155, 1053, 996, 957, 841
70				
15	NHR (6 ppm)* (): observation frequency	[270HHz] 1.70-1.93(6H,m), 2.10(2H,m), 2.48 (2H,t,J=6.9Hz), 3.01(2H,m), 3.21 (1H,m), 3.51(2H,t,J=6.6Hz), 3.93 (2H,s), 4.18(2H,m), 4.29(2H,m), 7.14(2H,t,J=8.6Hz), 7.46-7.63 (3H,m), 7.79(1H,m), 7.97(2H,dd, J=5.3Hz,8.6Hz)	(270Mhz) 1.65-1.95(6H.m), 2.08(2H.m), 2.47(2H.m), 2.97(2H.m), 3.09(1H, m), 3.50(2H.t.)=6.6hz), 3.92(2H, s), 4.17(2H.m), 4.29(2H.m), 6.82- 7.00(2H.m), 7.49-7.50(3H.m), 7.76-7.88(2H.m)	(400HHz) 2.02-2.07(4H,m), 2.26(2H,m), 2.72(2H,t,J=6.6Hz), 3.05-3.15 (3H,m), 3.62(2H,t,J=6.6Hz), 4.03 (2H,s), 4.17(2H,m), 4.30(2H,m), 7.05(1H,m), 7.24(1H,m), 7.50- 7.59(3H,m), 7.67(1H,dd,J=5.1Hz, 8.7Hz), 7.79(1H,dd,J=5.1Hz,
20 g	C	[270HHz] 1.70-1.93(6H.m (2H.t.)-6.9Hz] (1H.m), 3.51(2 (2H.s), 4.18(2 (2H.s), 4.2H(2 (3H.m), 7.79(1)	(270HHz) 1.65-1.95(6H.m), 2.9 2.47(2H.m), 2.9 1), 3.50(2H.t.), 4.17(2H.m), 7.4 7.00(2H.m), 7.4	442) (2H, t., J (2H, t., J s), 4.1 (3H, m),
ont		(270MHz) 1.70-1.9 (2H,t,J= (1H,m), (2H,s), 7.14(2H, (3H,m), 2.34(2H,)=5.3Hz,	(270HHz) 1.65-1.9 2.47(2H. m). 3.50 s). 4.17 7.00(2H.	(400Mtz) 2.02-2.0 2.72(2H, (3H,m), (2H,s), 7.05(1H, 7.59(3H, 8.7Hz),
25				٨.
S S S S S S Table 1 (Cont'd)	Property Melting point {recrystalli- zation solvent}	Colorless prism crystals 115.0-116.0°C (ethanol)	Pale yellow oil	Colorless oil [monofumarate] Colorless powdery crystals 135.5-137.0°C [ethanol]
35		ÿ_=°		Ö
40	Structural		, (cH ₀), (- (CH ₂₁ - N
45		, X in o	C S S S S S S S S S S S S S S S S S S S	Cox in o
50	Comp'd No.	09	7	24

" Measured in CDCl $_3$ with IMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method	(film) 2970, 1613, 1328, 1268, 1155, 1121, 1053, 956, 768, 746	(film) 2945, 2885, 2818. 1492, 1331, 1262. 1155, 1052, 949. 772, 738, 693	(KBr) 2952, 2360, 1628, 1518, 1471, 1444, 1320, 1226, 1156, 1054, 1014, 956, 830, 766, 737, 706, 685, 668
10			"		42).
15		NHR (6 ppm)* (): observation frequency	(4004Hz) 1.86(2H,m), 2.03-2.08(4H,m), 2.16 (2H,m), 2.51(2H,t,J=7.1Hz), 3.03- 3.10(3H,m), 3.53(2H,t,J=6.8Hz), 3.92(2H,s), 4.17(2H,m), 4.29(2H,s), 7.05(1H,dt,J=2.0Hz,8.8Hz), 7.23(1H,dd,J=2.0Hz,8.6Hz), 7.50- 7.23(1H,dd,J=2.0Hz,8.6Hz), 7.50- 7.58(3H,m), 7.68(1H,dd,J=5.1Hz,8.7Hz), 7.79(1H,dd,J=1.7Hz,J=8.9Hz)	(2704Hz) 1.89(2H,quint,J=6.9Hz), 2.56 [2H,t,J=6.9Hz), 2.70(4H,m), 3.50-3.61(6H,m), 3.94(2H,s), 4.19 (2H,m), 4.29(2H,m), 7.36(1H,t,J= 7.6Hz), 7.43-7.62(4H,m),7.77- 7.85(2H,m), 7.91(1H,d,J=8.6Hz)	(400HHz) 1.90(2H.quint, J=6.9Hz), 2.07(4H, m), 2.19(2H,m), 2.53(2H,t,J=6.9Hz), 2.99-3.15(3H,m), 3.53(2H,t,J=6.9Hz), 3.53(2H,t,J=6.9Hz), 3.94(2H,s), 4.18(2H,m), 4.29(2H,m), 6.90(1H,dt,J=2.1Hz,8.9Hz), 7.08(1H,dd,J=2.1Hz,8.9Hz), 7.80(3H,m), 7.71(1H,dd,J=5.1Hz,8.9Hz), 8.9Hz), 7.80(3H,m), 9.79(3H,br.s)
20	נים)		400HHz) .86(2H, 1) 2H, m), 1.10(3H, 1) 1.92(2H, 1) 1.92(3H, 1) 7.28(3H, 1) 3.7Hz),	(270MHz) 1.89(2H, c)= (2H, t, J= 3.50-3.6 (2H, m), 7.6Hz), 7.85(2H,	(400MHz) 1.90(2H, 1.90(2H, 1.90-3.1 2.99-3.1 3-6.9Hz) 4.29(2H, 7.08(1H, 7.08(3H, 8.9Hz),
(Con		(40 (2h 3.1 3.1 3.9 3.9 7.6 7.6	(2) (2) (3) (3) (4) (7)	2. E . S . C . C . C . C . C . C . C . C . C
25	7		_		E
<i>30</i>	Table 1 (Cont'd)	Property Helting point (recrystalli- zation solvent)	Colorless oil [monofumarate] Colorless prism crystals 174.5-176.5°C (ethanol)	Pale yellow oil	Colorless prism crystals 169.5-171.5°C (chloroform. -ethyl ether)
35		- a		= z	
40		Structural	() - () -	0 ×	(CH ₂),
4 5					
50		Comp`d No.	8	4	45

* Measured in CDCl₃ with TMS as an internal standard unless otherwise specifically indicated.

5	IR (cm ⁻¹) (): measuring method	(KBr.) 2955, 2888, 2804, 1594, 1482, 1432, 1324, 1240, 1158, 1054, 953, 766, 690	(KBr) 2954, 2905, 2810, 1585, 1548, 1497, 1360, 1323, 1160, 1054, 982, 956, 762, 690	(film) 2945, 2808, 1682, 1597, 1506, 1308, 1160, 976, 852, 736
ont'd)	NHR (6 ppm)* (): observation frequency	(270MHz) 1.87(2H,m), 2.50(2H,t,J=7.3Hz), 2.56(4H,m), 3.47-3.58(6H,m), 3.92(2H,s), 4.17(2H,m), 4.27 (2H,m), 6.58-6.68(2H,m), 7.43- 7.51(4H,m), 7.80(1H,d,J=7.3Hz), 8.18(1H,d,J=4.0Hz)	(2704Hz) 1.87(2H.quint, J=6.9Hz), 2.44- 2.59(6H,m), 3.53(2H,t,J=6.9Hz), 3.82(4H,m), 3.92(2H,s), 4.17 (2H,m), 4.28(2H,m), 6.48(1H,t,J=4.6Hz), 7.47-7.53(3H,m), 7.79 (1H,m), 8.30(2H,d,J=4.6Hz)	(400HHz) 1.75-1.96(6H.m), 2.12(2H.m), 2.51(2H,t,J=7.0Hz), 3.01(2H.m), 3.21(1H.m), 3.49(2H.m), 3.57- 3.70(4H.m), 4.13(2H.s), 7.14(2H,t,J=8.6Hz), 7.39(1H,m), 7.50(1H,m), 7.76(1H,dd,J=1.2Hz,7.9Hz), 7.91-8.01(3H,m)
Table 1 (Cont'd)	Property Melting point (recrystalli- zation solvent)	Colorless needle crystals 122.0-122.5°C (ethy) acetate -hexane)	Pale yellow prism crystals 118.5-119.5°C (ethyl acetate -hexane}	Brown of 1
35	ura) ula	S, '' (CH ₂₎ , '' (CH ₂), ''	0 × 10 × 10 × 10 × 10 × 10 × 10 × 10 ×	
45	Structural	S. (CH ₂)	Cox 20 10 10 10 10 10 10 10 10 10 10 10 10 10	Negation 1. So
50	Comp. d	94	43	æ

 * Measured in CDCl $_{
m J}$ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm^{-1}) (): measuring method	(KBr.) 3622, 3538, 2921, 1626, 1471, 1314, 1162, 1070, 828, 770	(KBr) 2930, 2829, 1569, 1515, 1296, 1172, 1156, 918, 826, 780	(KBr) 3425, 2942, 2820, 1592, 1514, 1456, 1312, 1232, 1163, 822, 759
10			13 004z. 7.39	2	13 87 76
15	Table 1 (Cont'd)	NMR (6 ppm)* (): observation frequency	(400MHz) 1.98(2H,quint,J=7.0Hz), Z.09(4H,m), 2.25(2H,quint,J=7.0Hz), Z.09(4H,m), 2.25(2H,quint,J=7.0Hz), 3.03 -3.18(3H,m), 3.49(2H,m), 3.58-3.67, (4H,m), 4.13(2H,s), 6.89(1H,dt,J=2.0Hz,8.9Hz), 7.08(1H,dd,J=2.0Hz,8.9Hz), 7.36(1H,dd,J=1.2Hz,7.39,7.7), 7.7Hz), 7.72(1H,dd,J=5.1Hz,8.9Hz), 7.77(1H,d,J=7.7Hz), 7.95(1H,dd,J=1.7Hz,7.39,7.39,7), 7.77(1H,d,J=7.7Hz), 7.95(1H,d,J=7.7Hz)	(400MHz) [0MSO-d ₆ /TMS] 1.83(2H,quint,J=6.9Hz), 2.41(2H,t,J=6.9Hz), 2.50(4H,m), 2.95(4H,m), 3.45(2H,t,J=6.9Hz), 3.58(2H,m), 3.70(2H,m), 4.08(2H,s), 6.64(2H,d,J=8.9Hz), 5.76(2H,d,J=8.9Hz), 7.66(2H,d,J=8.9Hz), 7.66(1H,dt,J=1.0Hz,7.8Hz), 7.68(1H,dt,J=1.0Hz,7.8Hz), 7.99(1H,d,J=7.8Hz), 8.74(1H,s)	(400Hkz) 2.70(4H,m), 2.77(2H,t,J=6.6Hz), 3.13 (4H,m), 3.46(2H,m), 3.63(2H,m), 3.70(2H,t,J=6.6Hz), 4.21(2H,s), 6.87 (2H,m), 6.96(2H,m), 7.39(1H,t,J=7.5 Hz), 7.51(1H,dt,J=1.2Hz,7.5Hz), 7.76 (1H,d,J=7.5Hz), 7.94(1H,d,J=7.5Hz)
25	1 (77	ery	dery er)	E ye
30	Table	Property Melting point (recrystalli- zation solvent)	Colorless powdery crystals 144.0-144.5°C (chloroform -ethyl ether)	Colorless powdery crystals 173.0-175.0°C (acetonitrile-isopropyl ether)	Colorless prism crystals 124.0-124.5°C (ethy) acetate hexane)
35		[8	, z z z	F 0	S." (CH ₂₎₂ H N - F
40		Structural formula	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	\ \frac{3}{2}	N ₂ (4H)
45		x	S S. W. (CH ₂₎ - N.	S S (CHU) N. CONDINAL N. S.	s s s s s
50		Comp'd No.	64	S	15

 st Measured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

	79	_		
5	IR (cm ⁻¹) (): measuring method	(KBr.) 2944, 2829, 1514. 1439, 1302, 1240. 1161, 782, 754. 660	(KBr.) 2934, 2826, 1588, 1514, 1456, 1421, 1385, 1303, 1270, 1242, 1163, 1070, 1000, 974, 922, 818, 773, 764,	(KBr) 3425, 2942, 2822, 1592, 1514, 1456, 1312, 1232, 1163, 822, 759
•	IR (cm ⁻¹) measuring	2829, 1302, 782,	(KBr.) 2934, 2826, 1514, 1456, 1385, 1303, 1163, 1163, 1000, 974, 728, 656	2942, 1514, 1232, 759
	()	(KBr) 2944, 1439, 1161, 660	(KBr.) 2934, 1514, 1385, 1242, 1000, 818, 728,	(KBr) 3425. 1592. 1312. 822.
10	Vone), 3.09), 3.70(2H, 2H,m), c.8.0Hz), 76(1H,	H.m). 7.14.7 1.14.7 1.34z.	44.m), 2.76 3.7Hz, (2H,t, n), z.8.0Hz), 82(1H,dd,
15	MMR (6 ppm)* (): observation frequency	(400HHz) 2.69(4H,m), 2.78(2H,t,J-6.6Hz), 3.09 (4H,m), 3.45(2H,m), 3.62(2H,m), 3.70(2H, t.J-6.6Hz), 4.21(2H,s), 6.78(2H,m), 6.84(2H,m), 7.39(1H,dt,J-1.0Hz,8.0Hz), 7.50(1H,dt,J-1.3Hz,8.0Hz), 7.76(1H,dt,J-1.3Hz,8.0Hz), dd,J-1.3Hz,8.0Hz), 7.94(1H,dd,J-1.0Hz,8.0Hz)	(400HHz) 1.90(2H.quint,J=6.7Hz), 2.01(1H,m), 2.21(1H,m), 2.57(2H.t,J=6.7Hz), 2.64(4H,m), 2.80(2H.dt,J=3.7Hz,14.7 Hz), 3.10(4H,m), 3.17(2H,m), 3.67(2H,t,J=6.7Hz), 4.51(2H,s), 6.76(2H,m), 6.84(2H,m), 7.45(1H,dt,J=1.3Hz,7.8Hz), 7.84(1H,dt,J=1.3Hz,7.8Hz), 7.841z), 8.06(1H,dd,J=1.3Hz,7.8Hz)	(400MHz) 2.02(1H,m), 2.22(1H,m), 2.79(4H,m), 2.76 (2H,t,J=6.5Hz), 2.80(2H,dt,J=3.7Hz, 14.7Hz), 3.09-3.17(6H,m),3.72(2H,t, J=6.5Hz), 4.56(2H,s),6.76(2H,m), 6.84(2H,m), 7.44(1H,dt,J=1.0Hz,8.0Hz), 7.54(1H,dt,J=1.2Hz,8.0Hz), 7.82(1H,dd, J=1.2Hz,8.0Hz), 8.07(1H,dd,J=1.0Hz,8.0Hz)
20 (g (d)	H ()	(400Hz) 2.69(41,m), 2 (44,m), 3.45(1,J=6.6Hz), 4 6.84(24,m), 7 7.50(14,dt,J= dd,J=1.3Hz,8.	(400HHz) 1.90(2H.quint 2.21(1H.m). 2 2.64(4H.m). 2 2.64(4H.m). 3 1.37(2H.t.)=6 5.76(2H.m). 6 1.31.3Hz.7.8Hz 7.8Hz), 7.8Zf 7.8Hz), 8.06((400Hhz) 2.02(1H.m), 2 (2H.t.)=6.5Hz 14.7Hz), 3.09 J=6.6Hz), 4.5 6.84(2H.m), 7 7.54(1H.dt.)= J=1.2Hz,8.0Hz
25 0		8	es (ery)
S % Table 1 (Cont'd)	Property Melting point {recrystalli- zation solvent}	Colorless needle crystals 188.0-188.5°C· (ethyl acetate -hexane)	Pale yellow prism Crystals 183.0-184.5°C (acetonitrile- isopropyl ether)	Pale brown powdery crystals 178.0-179.0°C (acetonitrile- isopropyl ether)
35		# \\	HO OH	¥0
40	Structural	S S (CH _{2),N} (CH _{2),N} N	S S CHADAN N	S CCH LINE N
45	:			
50	Comp'd No.	25	æ	24

 ullet Measured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method	(film) 2947, 2823, 1506. 1456, 1329, 1232. 1146, 1062, 954, 826, 769, 688	(film) 1520, 1450, 1380, 1330, 1240, 1180, 1080, 790	(film) 1510, 1460, 1330, 1230, 1170, 1070, 950, 830, 760, 700
				m & 4	
15		ния (б ppm)* (): observation frequency	(270MHz) 1.88(2H,m), 2.52(2H,t,J=7.3Hz), 2.63(4H,t,J=5.3Hz), 3.13(4H,t,J=5.3Hz), 3.13(4H,t,J=6.6Hz), 3.26(6H,s), 3.54(2H,t,J=6.6Hz), 3.98(2H,s), 6.84-6.99 (4H,m), 7.48-7.59(2H,m), 7.68(1H,dd,dJ=2.0Hz,7.3Hz), 7.84(1H,dd,	(2704Hz) 1.83(2H,m), 2.39(2H,t,J=7.3Hz), 2.53 (4H,t,J=5.3Hz), 3.09(4H,t,J=5.3Hz), 3.69(2H,t,J=7.3Hz), 3.80(3H,s), 6.08 (1H,s), 6.83-7.00(4H,m), 7.54 (1H,dt,J=1.3Hz,7.9Hz), 7.63(1H,dt,J=1.3Hz,7.9Hz), 7.94[1H,dd,J=1.3Hz,7.9Hz), 7.9Hz), 7.86(1H,dd,J=1.3Hz,7.9Hz),	(2704Hz) 1.86(2H.m), 2.50(2H.m), 2.60 (4H,t,J=5.3Hz), 3.12(4H,t,J=5.3Hz), 3.38(2H,m), 3.54(3H,s), 3.85(1H,dd,J=4.6Hz,15.2Hz), 4.10(1H,dd,J=4.6Hz,15.2Hz), 4.39(1H,m), 6.84-6.99(4H,m), 7.44-7.59(3H,m), 7.83(1H,dd,J=1.3Hz,7.9Hz)
20	Table 1 (Cont'd)	NMR (): ob	(270MHz) 1.88(2N.m), 2.5 2.63(4H, t.J=5.3 5.3Hz), 3.26(6) 6.6Hz), 3.98(2) (4H,m), 7.48-7 dd,J=2.0Hz,7.39	(2704Hz) 1.83(2H.m), 2 (4H.t.,J=5.3Hz) 3.69(2H.t.,J=7 (1H,s), 6.83-7 (1H,dt.,J=1.3Hz) 1.3Hz,7.9Hz), 7.9Hz), 7.8E(1)	(2704Hz) 1.86(ZH.m), 2.50(ZH.m), 2.6 (4H,t,J=5.3Hz), 3.12(4H,t,J 5.3Hz), 3.39(ZH.m), 3.54(3H 3.85(1H,dd,J=4.6Hz,15.ZHz), 4.10(1H,dd,J=4.6Hz,15.ZHz), 4.39(1H,m), 6.84-6.99(4H,m) 7.44-7.59(3H,m), 7.83(1H,dd 1.3Hz,7.9Hz)
25) (
30	Table	Property Helting point (recrystalli- zation solvent)	Pale yellow of l	Pale yellow oil [dihydrochloride] Pale yellow powdery crystals (methanol -ethyl ether)	Pale yellow oil [dihydrochloride] Pale yellow powdery crystals 180.0-182.0°C (methanol -ethyl ether)
35			1	<u>.</u> 1	ц 1
40		Structural formula	CH ₃ O OCH ₃	och, s.', chij, M.	OCH, OCH,
50		Comp`d No.	SS	8	. 53

 st Heasured in CDCl $_{3}$ with IMS as an internal standard unless otherwise specifically indicated.

		_			
5	·	IR (cm ⁻¹) (): measuring method	(film) 2942, 2820, 1681, 1599, 1505, 1446, 1337, 1160, 975, 854, 760, 702	(film) 2943, 2821, 1515, 1506, 1456, 1335, 1232, 1170, 1069, 827, 763, 705	(film) 2948, 2827, 1520, 1463, 1337, 1228, 1163, 924, 824, 765, 727, 700
10					
15		KMR (6 ppm)* (): observation frequency	(270HHz) (21.m), 3.00(2H.m), 2.10(2H.m), 2.48 (2H.m), 3.00(2H.m), 3.20(1H.m), 3.32(1H.m), 3.42(1H.m), 3.55(3H.s), dd,J=4.6Hz, 15.2Hz), 4.09(1H, dd,J=4.6Hz, 15.2Hz), 4.38(1H,t.J= 4.6Hz), 7.14(2H.m), 7.45-7.60(3H.m), 7.83(1H,d,J=7.2Hz), 7.96(2H.m)	(270HHz) 1.29(3H,t,J=7.3Hz), 1.85 (2H,quint,J=7.3Hz), 2.39-2.70 (6H,m), 3.11(4H,m), 3.31(1H,m), 3.48(1H,m), 3.60-3.88(3H,m), 4.09 (1H,dd,J=4.6Hz,14.5Hz), 4.51 (1H,t,J=4.6Hz), 6.80-7.03(4H,m), 7.40-7.61(3H,m), 7.82(1H,d,J=7.3Hz)	(270HHz) (36(2H,m), 2.51(2H,m), 2.63 (4H,m), 3.08(4H,m), 3.33(1H,m), 3.43(1H,m), 3.52(3H,s), 3.83 (1H,dd,J=4.6Hz,15.2Hz), 4.37 (1H,dd,J=4.6Hz,15.2Hz), 4.37 (1H,t,J=4.6Hz), 6.73(2H,d,J= 3.6Hz), 6.82(2H,d,J=8.6Hz), 7.45- 7.60(3H,m), 7.83(1H,d,J=7.9Hz)
20	Table 1 (Cont'd)	иня (): ob	(270HHz) 1.70-1.90(6H,m) (2H,m), 3.00(2H 3.32(1H,m), 3.4 dd,J-4.6Hz, 15. 4.6Hz, 7.14(2H 7.83(1H,d,J-7.2)	(270HHz) 1.29(3H,t.)=7.3Hz), 1.85 (2H,quint.)=7.3Hz), 2.39-2.7 (6H,m), 3.11(4H,m), 3.31(1H,m), 3.48(1H,m), 3.60-3.88(3H,m), (1H,dd,)=4.6Hz,14.5Hz), 4.51 (1H,t.)=4.6Hz), 6.80-7.03(4H,t.)=4.6Hz), 7.82(1H,d.)	(270HHz) 1.86(2H,m), 2.3 (4H,m), 3.08(4) 3.43(1H,m), 3.3 (1H,dd,J=4.6Hz) (1H,t,J=4.6Hz) 8.6Hz), 6.82(2) 7.60(3H,m), 7.
25	٦ (
30	Table	Property Melting point (recrystalli- zation solvent)	Pale yellow oil	Pale yellow oil	Pale yellow oil
35		a a	- c c c c c c c c c c		¥ 0
40		Structural formula	осн, сены, м. сены, м. се	OC-74, CCH ₂₎₃ N N	OCH, S. (CHabiN N
50		Comp'd No.	88	65	99

* Measured in CDCl₃ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method	(film) 1500, 1455, 1334, 1232, 1169, 1068, 817, 762, 701	(KBr) 2820, 1510, 1456, 1330, 1232, 1158, 1051, 995, 953, 816, 762	(film) 2973, 2820, 1510, 1331, 1231, 1154, 1075, 958, 826, 763
10	·	uency	2H, 1.41 5.2Hz), 1.61 6.80- 7.41-7.60	Hz), (2H, -4.23 0-7.04(4H,	d, H, A-
15		NHR (6 ppm) (): observation frequency	(270MHz) 1.83(2H.quint, J=7.3Hz), 2.47(2H, m), 2.57(4H, m), 3.09(4H, m), 3.41 (2H.m), 3.09(4H, m), 3.41 4.10(1H, dd, J=4.6Hz, 15.2Hz), 4.10(1H, dd, J=4.6Hz, 15.2Hz), 4.61 [1H, t, J=4.6Hz, 7.1(1H, d, J=11.9Hz), 4.78(1H, d, J=11.9Hz), 4.78(1H, d, J=11.9Hz), 5.80- 7.04(4H, m), 7.30-7.41(5H, m), 7.41-7.60 (3H, m), 7.83(1H, dd, J=2.0Hz, 7.3Hz)	(270HHz) 1.86(2H.m), 2.51(2H.t.J=7.3Hz), 2.61(4H.m), 3.12(4H.m), 3.52(2H, t.J=6.6Hz), 3.92(2H.s), 4.06-4.23 (2H.m), 4.23-4.38(2H.m), 6.80-7.04(4H, m), 7.45-7.65(3H.m), 7.80(1H.m)	(2704Hz) 1.33(3H,d,J=5.9Hz), 1.41(3H,d, J=5.9Hz), 1.86(2H,m), 2.50(2H, m), 2.50(4H,m), 3.12(4H,m), 3.52(2H,m), 3.80-4.04(4H,m), 6.84-6.99(4H,m), 7.46-7.58(3H,m), 7.77(1H,d,J-7.2Hz)
20	ıt'd))	(270WHz) 1.83(2H.qu m), 2.57(2H.qu (2H.m), 3 4.10(1H.d (1H.t,J=4 11.9Mz), 7.04(4H.m	(270MHz) 1.86(2H, 2.61(4H, 1.3=6.6Hz (2H,m), 4.3=6.7+45	(270Hrz) 1.33(3H.c) 1.33(3H.c) 1.5.9Hz) m). 2.62((2H.m). 6.99(4H.r) 7.77(1H.c)
25	Table 1 (Cont'd)	Property Melting point (recrystalli- zation solvent)	Pale yellow oil [dihydrochloride] Colorless powdery crystals 209.0-210.0 °C (methanol -ethyl ether)	Colorless prism crystals 102.0-104.0°C (ethyl acetate -hexane)	Colorless oil
30		10"	2 0 0 0 0		_
35		Structural formula	och, C	S. (CH _{2),} M. M. C. F.	
40		Stru	£		S. "Cota
45		Comp d		29	2
50		5 -			

 st Measured in CDCl $_{
m 3}$ with TMS as an internal standard unless otherwise specifically indicated.

	4 5	40	35	30	25	20	15	10		5	
				Table 1 (Cont'd)	1 (Co	nt'd)			i		
		Structural formula		Property Helting point (recrystalli- zation solvent)		NMR (NHR (6 ppm)* (): observation frequency	J	1R): me.	iR (cm ⁻¹) measuring	IR (cm ⁻¹) (): measuring method
	ر مر کی م	S. (CH ₂₎ , (CH ₂),		Colorless prism crystals 119.0-120.0°C (ethyl acetate -hexane)	·	(270HHz) 1.56(1H.m), 1.91(2H.m), 2.32(1H.r) (2H.t, J=7.3Hz), 2.67(4H.m), 3.14 (4H.m), 3.47(2H.t, J=6.6Hz), 3.97 (4H.m), 4.40(2H.s), 6.80-7.05(4H 7.49(1H.ddd, J=1.3Hz, 7.3Hz), 7.76(11 J=1.3Hz, 7.3Hz, 7.3Hz), 7.76(11 J=1.3Hz, 7.3Hz, 7.8Hz), 7.76(11	(270HHz) 1.56(1H.m), 1.91(2H.m), 2.32(1H.m), 2.58 (2H.t.,J=7.3Hz), 2.67(4H.m), 3.14 (4H.m), 3.47(2H,t.,J=6.6Hz), 3.97-4.28 (4H.m), 4.40(2H.s), 6.80-7.05(4H.m), 7.49(1H.ddd,J=1.3Hz,7.3Hz,7.9Hz), 7.60(1H,ddd,J=1.3Hz,7.9Hz), 7.76(1H.dd,J=1.3Hz,7.9Hz), 7.76(1H.dd,J=1.3Hz,7.9Hz), 7.86(1H,dd,J=1.3Hz,7.9Hz)	:.58 ; 50(1H,	(KBr) 2818, 1509, 1323, 1226, 1139, 1076, 1016, 960, 828, 768	(KBr.) 1509, 1 1139, 1 960,	1323. 1076, 828.
· (_)		S. N. (CH ₂₎₃ N N		Colorless oil [dihydrochloride] Colorless powdery crystals 112 °C (decomposed) (chloroform-	ed)	(270Hhz) 1.87(2H,quint,J=6.9Hz), 2.39-2.70(3.12(4H,m), 3.28(1H,m), 3.32-3.47(2H,m), 3.60(1H,m), 4.00(1H,d,J=1:4.10(1H,d,J=5.3Hz,9.6Hz), 4.20(1H,d,J=15.8Hz), 4.61(1H,m), 6.80(4H,m), 7.57(1H,m), 7.61(1H,m), 7.61(1H,m), 7.61(1H,m), 7.61(1H,m), 7.61(1H,m), 7.91(1H,m), 7.80(1H,d,J=7.3Hz)	(2704Hz) 1.87(2H,quint,J=6.9Hz), 2.39-2.70(6H,m), 3.12(4H,m), 3.28(1H,m), 3.32-3.47 (2H,m), 3.60(1H,m), 4.00(1H,d,J=15.2Hz), 4.10(1H,dt,J=5.3Hz,9.6Hz), 4.20 (1H,d,J=15.8Hz), 4.61(1H,m), 6.80-7.03 (4H,m), 7.47(1H,m), 7.57(1H,m), 7.71 (1H,d,J=7.9Hz), 7.80(1H,d,J=7.3Hz)		(film) 2945, 2879, 2819, 1510, 1454, 1329, 1232, 1171, 1046, 950, 826, 762, 704	1454, 1 1171, 1 1171, 1	2819. 1329. 1046. 762.
	ر "کے یہ ہ	S X (CH ₀), 10	Ċ	Colorless prism crystals 137.5-138.5°C (ethyl acetate -hexane)		(270MHz) 1.92(2H,quint,J=7.3Hz), 2.54(2H,t 7.3Hz), 2.63(4H,m), 3.13(4H,m), 3.40-3.53(2H,m), 3.56-3.70(4H,m), 4.11(2H,s), 6.83-7.01(4H,m), 7.40 (1H,dt,J=1.3Hz,7.9Hz), 7.52(1H,dt J=1.3Hz,7.9Hz), 7.52(1H,d,J=1.3Hz,7.9Hz) 7.94(1H,d,J=1.3Hz,7.9Hz)	(27044z) 1.92(2H.quint, J-7.3Hz), 2.54(2H, t, J-7.3Hz), 2.54(2H, t, J-7.3Hz), 2.54(2H, t, J-7.3Hz), 3.13(4H, m), 3.40-3.53(2H, m), 3.56-3.70(4H, m), 4.11(2H, s), 6.83-7.01(4H, m), 7.40 (1H, dt, J-1.3Hz, 7.9Hz), 7.52(1H, dt, J-1.3Hz, 7.9Hz), 7.77(1H, d, J-7.3Hz), 7.94(1H, dd, J-1.3Hz, 7.9Hz)	<u>.</u>	(KBr) 2813, 1508, 1299, 1222, 1159, 1136, 1115, 828	1159, 828	.136,

 ullet Measured in CDC $_{
m J}$ with TMS as an internal standard unless otherwise specifically indicated.

		IR (cm ⁻¹) {): measuring method	2820. 1380. 1164. 941. 757 656	2821. 1309. 1072. 817. 719.	(film) 2929, 2819, 1511. 1455, 1378, 1311, 1231, 1164, 1072, 931, 826, 756, 717
5		IR (cm ⁻¹) measuring	1456. 1456. 1231. 1024. 779.	2944, 1454, 1163. 971, 742,	2819. 1 1378. 1 1164. 1 826.
		 F 96	(film) 3057, 2947, 2820, 1510, 1456, 1380, 1312, 1231, 1164, 1074, 1024, 941, 817, 779, 757 737, 716, 656	(f11m) 3058, 2944, 2821, 1509, 1454, 1309, 1235, 1163, 1072, 1012, 971, 817, 759, 742, 719, 656	(film) 2929. 1455. 1231. 931. 717
		_	21 3 . 11 . 112. 3		
10		frequency	.02(14,m), 2.), 2.64(44,m)), 3.10-3.22), 4.50(24,s) ,4.5(14,dt,J=1,2,7,7kz), 7.8	.71(4H,m), 2H,dt,J= 6H,m), 2H,s), 2H,s), 4S(1H,dt, tt,J=1.3Hz, tt,7.8Hz),	(2H,quint,), 2.63(4H,m) 6.9Hz), ,m), 7.42 .55(1H,dt, dd,J=1.3Hz, Hz,7.9Hz)
15		NHR (6 ppm)* (): observation frequency	(f11m) 1.90(2H,quint,J=6.7Hz), 2.02(1H,m), 2.21 3057, (1H,m), 2.57(2H,t,J=6.7Hz), 2.64(4H,m), 1510, 2.81(2H,dt,J=3.7Hz,14.6Hz), 3.10-3.22 1312, (6H,m), 3.67(2H,t,J=6.7Hz), 4.50(2H,s), 1074, 6.87(2H,m), 6.96(2H,m), 7.45(1H,dt,J=1.1Hz, 817, 7.7Hz), 7.54(1H,dt,J=1.3Hz,7.7Hz), 7.83 737, [H,dd,J=1.3H,7.7Hz), 8.07(1H,dc,J=1.1H,7.7Hz)	(400MHz) 2.03(1H.m), 2.24(1H.m), 2.71(4H.m), 2.75(2H,t,J=6.6Hz), 2.82(2H,dt,J= 3.7Hz,14.7Hz), 3.10-3.20(6H.m), 3.72(2H,t,J=6.6Hz), 4.56(2H,s), 6.87(2H,m), 6.96(2H,m), 7.45(1H,dt,J=1.0Hz,7.8Hz), 7.55(1H,dt,J=1.3Hz,7.8Hz), 7.83(1H,dd,J=1.3Hz,7.8Hz), 8.08(1H,dd,J=1.0Hz,7.8Hz)	[270HHz] 1.16(6H,t.J=7.3Hz], 1.90(2H,quint, J=6.9Hz), 2.42-2.70(6H,m), 2.63(4H,m), 3.13(4H,m), 3.62(2H,t.J=6.9Hz), 4.18(2H,s), 6.82-7.01(4H,m), 7.42(1H,dt,J=1.3Hz,7.9Hz), 7.55(1H,dt, J=1.3Hz,7.9Hz), 7.55(1H,dt, J=1.3Hz,7.9Hz), 7.95(1H,dt, J=1.3Hz,7.9Hz), 7.95(1H,dt, J=1.3Hz,7.9Hz), 7.95(1H,dt, J=1.3Hz,7.9Hz)
20	Table 1 (Cont'd)	Ö	(400 MHz) 1.90(2H,quin (1H,m), 2.57 2.81(2H,dt,J (6H,m), 3.67 6.87(2H,m), 7.7Hz), 7.55	(400MHz) 2.03(1H.m), 2.76(2H, t, J) 3.7Hz, 14.7H, 3.72(2H, t, J) 6.87(2H, m), J=1.0Hz, 7.8 7.8Hz), 7.8 8.08(1H, dd,	(270Hhz) 1.16(6H,t.J) 3-6.9Hz), 2 3.13(4H,m), 4.18(2H,s), (1H,dt,J=1,3Hz,7.5) 7.9Hz), 7.5
25	7	_	_		de]
	Table	Property Melting point (recrystalli- zation solvent)	Colorless oil [monomaleate] Colorless prism crystals 139.0-140.5°C (chloroform-	Colorless oil	Coloriess oil [dihydrochloride] Coloriess powdery crystals 183.0-166.5°C (methanol)
30		Pro Heli (rec	Colc (Colc 1339 eth	. · ·	
35	·		Ò		
40		Structural	S (CH ₁₀ , 10)	S CHANN N	S S S S S S S S S S S S S S S S S S S
			~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	r v v
45					
		Comp [*] d No.	69	82	25
50		I	1	1	1

 ullet Heasured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

Table 1 (Cont'd) Property P	5		IR (cm ⁻¹) (): messuring method	(KBr.) 1706, 1508, 1341, 1276, 1242, 1172, 1126, 816, 768	(CHCl ₃ solution) 2815, 1695, 1590. 1350, 1170	(KBr) 2947, 2772, 1705, 1680, 1598, 1505, 1339, 1225, 1170, 976, 854, 768
Table 1 (Cont'd Froperty Froperty	10					
Table 1 (Cont'd Froperty Froperty	15		MMR (5 ppm)*): observation frequency	int.J-6.9Hz), 2.47(2H.t., 2.57(4H.m), 3.11(4H.m), J-6.9Hz), 4.48(2H.s), 4H.m), 7.68-7.92(3H.m), J-1.3Hz,7.3Hz)	int, J=6.9Hz), 2.46 9Hz), 2.57(4H,m), 3.18 29(2H,t,J=6.9Hz), 6.85(1H,t,J=7.3Hz), J=7.9Hz), 7.26(2H,m), 3H,m), 8.07(1H,dd,	6H,m), 2.08(2H,m), J=7.3Hz), 2.93(2H,m), , 3.26(2H,t,J=7.3Hz), , 7.13(2H,m), 7.70- , 7.66(1H,m), 7.95 O7(1H,dd,J=1.3Hz,
Table 1 (Co Tytals Tomula Table 1 (Co Tytals Tomula	20	(p		Hz), Hz), ZH.t., 7.02(Hz) 2H.qu ,J=6. (), 3. 2H,d, 2H,d, 7.93(Hz) 1.90(2H,t. 2H,t) 1H,m) 2H,s) 2H,s)
Properties Pro		(Cont		(270th 1.81(2 3.6.9) 3.29(2 6.80-) 8.07(1)	(270HI) 1.81() (2H, t (4H, m 4.48() 5.92() 7.68-	(270H 1.70- 2.42(3.18(4.46(7.83((2H.m. 7.9Hz
Properties Pro	25		a			
27 CH ₂), N N N N N N N N N N N N N N N N N N N	30	Table	Property Melting point (recrystalli- zation solven	Brown needle crystals 110.0-112.0°C (?-propanol)	Coloriess oil	Brown needle crystals 109.0-110.0°((ethanol)
45 P. damo. 20 T. 22	35		, , , , , , , , , , , , , , , , , , ,			;=°
45 P. damo. 20 T. 22	40		uctur	ا الم	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	N.tg.
70 No. 72			Str	o=\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	0 = \(\frac{1}{2} \)	0 = z', o o
5, 7 '			ق ن ق ف	92	12	22
50	50		် ဦ =			

 st Measured in CDCl $_{
m 3}$ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): messuring method	(KBr.) 2830, 1600, 1496, 1347, 1248, 1174, 1126, 970, 923, 755	(KBr) 2943, 1684, 1597, 1508, 1340, 1242, 1176, 1156, 972, 840, 752	(KBr.) 3676, 2814, 2362, 1734, 1561, 1506, 1342, 1210, 1174, 970, 830, 784, 768
10	İ				
15		NMR (6 ppm)* (): observation frequency	(27044z) 1.92(2H,m), 2.63(2H,m), 2.75 (4H,m), 3.08(2H,t,J=6.9Hz), 3.35(4H,m), 4.69(2H,s), 6.89 (1H,t,J=7.3Hz), 6.95(2H,d,J= 8.0Hz), 7.22-7.33(2H,m), 7.38- 7.52(2H,m), 7.80(1H,m), 7.90 (1H,m), 13.00(1H,br.s)	[270HHz] [DHSO-d ₆ /THS] 1.40-1.81(6H,m), 2.02(2H,dt,J= 1.3Hz,9.9Hz), 2.30(2H,t,J=6.6Hz), 2.84(2H,m), 2.97(2H,t,J=6.6Hz), 3.20-3.42(1H,m), 4.59(2H,s), 7.33(2H,t,J=8.9Hz), 7.58-7.76 (2H,m), 7.79(1H,dd,J=1.3Hz,7.6Hz), 7.97-8.12(3H,m), 12.28(1H,s)	(2704Hz) 1.91(2H.m), 2.62(2H.m), 2.75 (4H.m), 3.08(2H,t,J=6.6Hz), 3.26(4H.m), 4.68(2H,s), 6.84- 7.03(4H.m), 7.47(2H.m), 7.81 (1H,dd,J=2.0Hz,7.3Hz), 7.89 (1H,dd,J=2.0Hz,7.3Hz), 12.72 (1H,br.s)
20		* ::	m), 2 3.08(m), 4 7.34z 7.22- 7.22- 13.00	(DHS), (BHZ), (B	.m). 3.08 .m). .m). .m).
25	Table 1 (Cont'd)		(270HHz) 1.92(2H, 1.92(2H, 1.92(4H, 1.92) 3.35(4H, 1.92 (1H, t.92(1H, 1.92) 7.52(2H, 1.93) (1H, 1.93)	(270Hhz) 1.40-1.8 1.3Hz,9. 2.84(2H, 3.20-3.4 7.33(2H, m). 7.97-8.1	(270Mkz) 1.91(2H.m (4H.m), 3 3.26(4H.m 7.03(4H.m (1H,dd,J= (1H,dd,J= (1H,br.s)
25	e 1	ਜ਼	d]e		. c .
30	Tabl	Property Melting point (recrystalli- zation solvent)	Colorless needle crystals 191.0-192.5°C (decomposed) [ethanol]	Colorless needle crystals 186.5-188.0°C (decomposed) (acetonitrile)	Coloriess prism crystals 173.0-174.0°C (ethyl acetate -hexane)
35					
40		Structural	NOH S. Y. (CH3) N. S.	NOH S. (CH2), N. C.	NOH S. N. (CH3), N. H.
45					
50		Comp d	73	74	27

* Measured in CDCl $_3$ with TMS as an internal standard unless otherwise specifically indicated.

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	IR (cm ⁻¹)	1673, 1234, 1059, 840,	1323. 922	1621, 1318, 1069, 828,
5	IR (cm ⁻¹) measuring	(KBr) 2963, 2815, 1673, 1593, 1317, 1234, 1208, 1154, 1059, 969, 853, 840, 763	(KBr.) 2043, 1508, 1323 1241, 1169, 922 820, 714	(film) 2948, 2818, 1621, 1505, 1455, 1318, 1233, 1176, 1069, 960, 922, 828, 779, 752
	= E	(KBr.) 2963, 1593, 1208, 969,	(KBr) 2843. 1241. 820.	(film) 2948, 1505, 1233, 960, 779,
10			. 걒	
15	NMR (6 ppm)** (): observation frequency	3-2.10(6H,m), 0-3.18(3H,m), 4.8Hz), 3.64 =2.9Hz,14.8Hz), 7.11(2H,m), 0(2H,m),	m). 2.46(1H.m), 4-3.01(4H.m), 4.5Hz), 3.60(1H, 3.86(1H,dt,J=7.3 J=2.9Hz,14.9Hz), 5.90(1H,br.s), 17-7.01(2H,m),	(t.)=6.7Hz), m), 3.84(2H, d,)=7.9Hz), 6.84-6.87 (m), 7.35(1H, m), 7.57(1H,
20	rty g point NHR (6 pm)* stalli- (): observation solvent)	(270MHz) 1.30-1.60(2H,m), 1.63-2.10(6H,m), 2.30-2.62(3H,m), 2.90-3.18(3H,m), 3.61(1H,dd,J-2.9Hz,14.8Hz), 4.65(1H,t,J-2.9Hz), 7.11(2H,m), 7.44(1H,m), 7.50-7.70(2H,m), 7.81(1H,m), 7.84-7.98(2H,m),	[270Mhz] 1.85(2H.m), 2.27(2H.m), 2.46(1H.m), 2.50-2.66(3H.m), 2.74-3.01(4H.m), 3.10(1H,dt,J=4.6Hz,14.5Hz), 3.60(1H,dd,J=2.7Hz,14.5Hz), 3.86(1H,dt,J=7.3Hz,13.9Hz), 4.24(1H,dd,J=2.9Hz,14.9Hz), 4.63(1H,t,J=2.9Hz), 5.90(1H,br.s), 6.65-6.84(2H.m), 6.87-7.01(2H.m), 7.38-7.64(3H.m), 7.83(1H,d,J=7.3Hz)	(400MHz) 1.96(2H,m), 2.45(2H,t,J=6.7Hz), 2.57(4H,m), 3.10(4H,m), 3.84(2H,t,J=6.7Hz), 6.66(1H,d,J=7.9Hz), 6.66(1H,d,J=7.9Hz), 6.84-6.87 (2H,m), 6.93-6.97(2H,m), 7.35(1H,d,J=7.8Hz), 7.91(1H,d,J=8.0Hz)
25	1			-
30 E T	Property Melting point (recrystalli-	Colorless powdery crystals 161.0-162.5°C (ethyl acetate)	Colorless plate crystals 170.0-174.5°C (decomposed) (ethyl acetate)	Coloriess oil
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40	Structural	0H 0, (CH ₃), N C (S)	oh S.'N (CH ₂₎₃ N N	S. N. (CHJ) N. N.
45		H - ().20	g — ;; ç,	
50	Comp. d	9/2	n .	78

 * Measured in CDCl $_{
m 3}$ with TMS as an internal standard unless otherwise specifically indicated.

5		IR (cm ⁻¹) (): measuring method
15		HMR {6 ppm}* (): observation frequency
20	Cont'd)	30 :()
25	Table 1 (Cont'd)	Property Helting point (recrystalli-
35		Pro Heli
40		Structural
45		b de .

IR (cm ⁻¹) (): measuring metho	(KBr.) 2944, 2804, 1513, 1477, 1459, 1330, 1240, 1155, 1012, 929, 822, 785, 766, 712
NHR (6 ppm)* (): observation frequency	(400MHz) 1.86(2H.m), 2.50(2H.t,J=7.1Hz), 2.60(4H.m), 3.01(2H.t,J=6.3Hz), 3.11(4H.m), 3.26(2H.t,J=7.0Hz), 3.90(2H.t,J=6.3Hz), 6.87(2H.m), 6.95(2H.m), 7.22(1H.d,J=7.7Hz), 7.37(1H.m), 7.44(1H.m), 7.83(1H,d,J=7.8Hz)
Property Helting point (recrystalli- zation solvent)	Colorless powdery crystals 103.5-104.0°C (ethyl acetate -hexane)
Structural formula	S. '(Cra), W. H.
Comp'd No.	79

 $^{\circ}$ Measured in CDCl $_{3}$ with TMS as an internal standard unless otherwise specifically indicated.

55 Test

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With respect to the compounds of the present invention, their anti-serotonin (5-HT) action and anti- α 1 action were investigated by the methods which will be described below. The results of some representative

compounds are shown in Table 2.

(1) Anti-serotonin action (anti-5-HT action)

The superior mesenteric artery of each Hartley male guinea pig (body weight: 300-500 g) was excised. A preparation cut in a helical form was suspended under resting tension of 0.3 g in a Magnus cylinder filled with the Tyrode solution which had been aerated with a gas mixture of 95% O₂ and 5% CO₂ and maintained at 37 °C. Using an isometric transducer ("UL-10", manufactured by SHINKOH K.K.) and a pressure preamplifier ("DSA-605A", manufactured by SHINKOH K.K.), variations in tension were measured. The isometric tensions were recorded on a pen-writing recorder ("VP-6537A", manufactured by NATIONAL K.K.). Taking the contraction induced by 10⁻⁵ M serotonin (5-HT) as 100%, the percent contractions in the presence of each test drug at 10⁻⁷ and 10⁻⁶ M were determined as anti-5-HT action.

(1) Anti-a₁ action

The thoracic aorta of each Hartley male guinea pig (body weight: 300-500 g) was excised. A preparation cut in a helical form was suspended under 1 g load in a Magnus cylinder filled with the Tyrode solution which had been aerated with a gas mixture of 95% O_2 and 5% CO_2 and maintained at 37 °C. Using an isometric transducer ("TB-612J", manufactured by NIHON KOHDEN) and a pressure preamplifier ("AP-620G", manufactured by NIHON KOHDEN), variations in tension were measured. The isometric tensions were recorded on a thermal pen-writing recorder ("WT-647G", manufactured by NIHON KOHDEN).

Taking the tonic contraction induced by 10^{-5} M norepinephrine (NE) as 100%, the percent contractions upon addition of each test drug at 10^{-8} and 10^{-7} M were determined as anti- α_1 action.

(Results)

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Table 2

Comp'd No.	Anti 5-HT action	n (% of Control)	Anti a1 action (% of Control)	
	10 ^{−7} M	10 ^{−6} M	10 ⁻⁸ M	10 ^{−7} M
35	38.3	7.5	99.0	89.3
37	15.7	8.3	100	96.2
40	37.4	13.4	98.3	73.2
46	39.8	5.5	100	97.4
55	23.4	7.4	100	89.2
56*	60.8	16.9	95.4	69.5
57 *	22.0	10.0	97.3	81.6
62	14.1	3.3	98.1	53.1
64	24.8	13.8	97.3	84.9
66	66.6	2.8	100	86.5
76	53.9	21.7	100	88.2
77	45.7	18.2	100	79.6

^{*} The compound in the form of a dihydrochloride was used as the test compound.

industrial Applicability

The benzothiazine derivatives (I) and their salts according to the present invention have strong serotonin-2 blocking action, have excellent selectivity to α_1 blocking action and have high safety. Accordingly, the present invention has made it possible to provide pharmaceuticals making use of antagonistic action against serotonin-2 receptors, for example, therapeutics for various circulatory diseases such as ishemic heart diseases, cerebrovascular disturbance and peripheral circulatory disturbance.

Claims

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1. A benzothiazine derivative represented by the following formula (I):

wherein the dashed line indicates the presence or absence of a bond, and when the dashed line indicates the presence of the bond, Z represents one of the following groups:

in which R₁ represents a substituted or unsubstituted alkyl group or a substituted or unsubstituted aralkyl group but, when the dashed line indicates the absence of the bond, Z represents one of the following groups:

in which R_2 represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, R_3 represents a hydrogen atom, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, R_4 represents a hydrogen atom, a substituted or unsubstituted alkyl group or a substituted or unsubstituted aralkyl group, K_1 , K_2 and K_3 each independently represents an oxygen atom or a sulfur atom, K_3 represents an ethylene group with one or more of the hydrogen atoms thereof optionally substituted by a like number of halogen atoms and/or alkyl, aryl, aralkyl and/or alkylidene groups or a trimethylene group with one or more of the hydrogen atoms thereof optionally substituted by a like number of halogen atoms and/or alkyl, aryl, aralkyl or alkylidene groups,

A represents a substituted or unsubstituted alkylene group, a substituted or unsubstituted alkenylene group or a substituted or unsubstituted alkynylene group,

Y represents CH, C = or a nitrogen atom; and, when Y represents CH, m stands for 0 or 1, n stands for 1 or 2, B represents an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an alkylene group, an alkenylene group, a substituted or unsubstituted hydroxymethylene group, a group -CHR $_5$ - in which R $_5$ represents a substituted or unsubstituted alkyl group, a

substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, or a substituted or unsubstituted cyclic or acyclic acetal group, when Y represents C = , m stands for 1, n stands for 1 or 2, B represents a group:

_₹

in which the double bond is linked to Y, R₆ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, but when Y represents a nitrogen atom, m stands for 0 or 1, n stands for 2 or 3, B represents a carbonyl group, a sulfonyl group, an alkylene group, an alkenylene group, or a group -CHR₇-in which R₇ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group,

E₁ and E₂ each independently represents a hydrogen atom or a lower alkyl group and,

D represents a substituted or unsubstituted aromatic hydrocarbon group or a substituted or unsubstituted aromatic heterocyclic group; or a salt thereof.

20 2. A benzothiazine derivative or a salt thereof according to claim 1, wherein in the formula (I), Z represents a group:

 x_2 x_3

wherein G, X₂ and X₃ have the same meanings as defined above.

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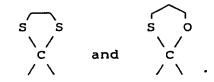
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A benzothiazine derivative or a salt thereof according to claim 1, wherein in the formula (I), Z represents one of groups:

o o C and C

4. A benzothiazine derivative or a salt thereof according to claim 1, wherein in the formula (I), Z represents one of groups:



55 5. A benzothiazine derivative or a salt thereof according to claim 1, wherein in the formula (I), Z represents a group:

wherein R4 has the same meaning as defined above.

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6. A benzothiazine derivative or a salt thereof according to claim 1, wherein in the formula (I), Z represents a group:

wherein R₁ has the same meaning as defined above.

- A benzothiazine derivative or a salt thereof according to any one of claims 1, 2, 3, 4, 5 and 6, wherein
 in the formula (I), A represents an ethylene or trimethylene group.
 - 8. A benzothiazine derivative or a salt thereof according to any one of claims 1, 2, 3, 4, 5, 6 and 7, wherein in the formula (I), Y represents CH, n stands for 2, B represents a carbonyl group, m stands for 0 or 1, and D represents a substituted or unsubstituted phenyl group.
 - A benzothiazine derivative or a salt thereof according to any one of claims 1, 2, 3, 4, 5, 6 and 7, wherein in the formula (I), Y represents a nitrogen atom, n stands for 2, m stands for 0, and D represents a substituted or unsubstituted phenyl group.
 - 10. A benzothiazine derivative or a salt thereof according to any one of claims 1, 2, 3, 4, 5, 6, 7, 8 and 9, wherein in the formula (I), E₁ and E₂ each represents a hydrogen atom.
 - 11. A process for the preparation of a benzothiazine derivative represented by the following formula (la):

wherein the dashed line indicates the presence or absence of a bond and, when the bond indicated by the dashed line is present,

Z₁ represents a group:



in which R_1 has the same meaning as defined above but, when the bond indicated by the dashed line is absent, Z_1 represents one of the groups represented by the following formulas:

$$R_2X_1$$
 X_1R_2 X_2 X_3 O NOR'₃ H OR'₁

C , C , C and C

in which R'₁ represents a substituted or unsubstituted alkyl group, R'₃ represents a substituted or unsubstituted aryl group or a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, and G, R₂, X₁, X₂ and X₃ have the same meanings as defined above,

and A, B, D, E₁, E₂, Y, m and n have the same meanings as defined above, which comprises: reacting a compound, which is represented by the following formula (II):

wherein Z_1 has the same meaning as defined above, with a compound represented by the following formula (III):

W-A-W' (III)

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wherein A has the same meaning as defined above and W and W' may be the same or different and individually represent a substituent easily replaceable with an amino group, to obtain a compound represented by the following formula (IV):

wherein A, W and Z₁ have the same meanings as defined above; and

then reacting the compound of the formula (IV) with a nitrogen-containing compound represented by the following formula (V):

$$(CH_2)_n$$

$$HN \qquad Y-(B)_m-D \qquad (V)$$

$$E_1 \qquad E_2$$

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

12. A process for the preparation of a benzothiazine derivative represented by the following formula (la):

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wherein A, B, D, E₁, E₂, Y, Z₁, m and n have the same meanings defined above, which comprises: reacting a compound, which is represented by the following formula (II):

wherein Z₁ has the same meaning as defined above, with a nitrogen-containing compound represented by the following formula (VI):

$$\begin{array}{c}
\begin{pmatrix}
CH_2 \\ n \\
V - (B)_m - D
\end{pmatrix}$$

$$\begin{array}{c}
V = (B)_m - D
\end{array}$$
(VI)

wherein A, B, D, E₁, E₂, W, Y, m and n have the same meanings as defined above.

13. A process for the preparation of a benzothiazine derivative represented by the following formula (lc):

$$\begin{array}{c}
O \\
S \\
O_2
\end{array}$$

$$\begin{array}{c}
A-N \\
E_1
\end{array}$$

$$\begin{array}{c}
Y-(B)_m-D
\end{array}$$
(Ic)

wherein A, B, D, E_1 , E_2 , Y, m and n have the same meanings defined above, which comprises converting Z_2 of a benzothiazine derivative to a carbonyl group, said derivative being represented by the following formula (lb):

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 & & \\$$

wherein Z₂ represents a group:

$$R_2X_1$$
 X_1R_2 or C

in which G, R_2 , X_1 , X_2 and X_3 have the same meanings as defined above, and A, B, D, E_1 , E_2 , Y, m and n have the same meanings as defined above.

14. A process for the preparation of a benzothiazine derivative represented by the following formula (le):

wherein, when Y represents CH, B' represents an oxygen atom, a sulfur atom, a sulfinyl group, a sulfonyl group, an alkylene group, an alkenylene group, a substituted or unsubstituted hydroxymethylene group, a group -CHR₅- in which R₅ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted cyclic or an acyclic acetal group, when Y represents a group C=, B' represents a group:

in which the double bond is linked to Y, R₆ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, but when Y represents a nitrogen atom, B' represents a carbonyl group, a sulfonyl group, an alkylene group, an alkenylene group or a group -CHR₇- in which R₇ represents a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group or a substituted or unsubstituted aralkyl group, and A, D, E₁, E₂, R₃, m and n have the same meanings as defined above, which comprises reacting a benzothiazine derivative, which is represented by the following formula (Id):

wherein A, B', D, E_1 , E_2 , Y, m and n have the same meanings as defined above, with a hydroxylamine represented by the following formula (VII):

NH₂OR₃ (VII)

wherein R₃ has the same meaning as defined above or with a derivative thereof.

15. A process for the preparation of a benzothiazine derivative represented by the following formula (If):

$$\begin{array}{c}
OH \\
S \\
O_2
\end{array}$$

$$\begin{array}{c}
(CH_2)_n \\
Y - (B')_m - D
\end{array}$$
(If)

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wherein A, B', D, E₁, E₂, Y, m and n have the same meanings as defined above, which comprises reducing a benzothiazine derivative represented by the following formula (Id):

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wherein A, B', D, E₁, E₂, Y, m and n have the same meanings as defined above.

16. A process for the preparation of a benzothiazine derivative represented by the following formula (lg):

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NOR₃

$$S_{O_2} \xrightarrow{(CH_2)_n} Y^{-(B)_m-D}$$
(Ig.

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wherein A, B, D, E₁, E₂, R₃, Y, m and n have the same meanings as defined above, which comprises: reacting a compound, which is represented by the following formula (VIII):

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wherein A and W have the same meanings as defined above, with a hydroxylamine represented by the following formula (VII):

NH₂OR₃ (VII)

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wherein R₃ has the same meaning as defined above or with a derivative thereof to obtain a compound represented by the following formula (IX):

wherein A, R₃ and W have the same meanings as defined above; and then

reacting the compound of the formula (IX) with a nitrogen-containing compound represented by the following formula (V):

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

17. A process for the preparation of a benzothiazine derivative represented by the following formula (Ih):

$$\begin{array}{c}
OH \\
S \\
O_2
\end{array}$$

$$\begin{array}{c}
(CH_2)_n \\
Y-(B)_m-D
\end{array}$$
(Ih)

wherein A, B, D, E_1 , E_2 , Y, m and n have the same meanings as defined above, which comprises reducing a compound, which is represented by the following formula (VIII):

wherein A and W have the same meanings as defined above, to obtain a compound represented by the

following formula (X):

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OH SN A-W (X)

wherein A and W have the same meanings as defined above, and then reacting the compound of the formula (X) with a nitrogen-containing compound represented by the following formula (V):

$$(CH2)n$$
HN $Y-(B)_m-D$ (V)

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

18. A process for the preparation of a benzothiazine derivative represented by the following formula (li):

$$\begin{array}{c}
 & \text{OR}_{8} \\
 & \text{SN} \\
 & \text{N} \\
 & \text{O2}
\end{array}$$

$$\begin{array}{c}
 & \text{CH}_{2})_{n} \\
 & \text{Y-(B)}_{m}-D
\end{array}$$
(Ii)

wherein R₈ represents a substituted or unsubstituted alkyl group or a substituted or unsubstituted aralkyl group, and A, B, D, E₁, E₂, Y, m and n have the same meanings as defined above, which comprises:

reacting a compound, which is represented by the following formula (X):

50 OH
$$S N A - W$$
 $O(X)$

wherein A and W have the same meanings as defined above, with a compound represented by the following formula (XI):

R₈-W" (XI)

wherein W" represents an eliminative substituent and R₈ has the same meaning as defined above, to obtain a compound represented by the following formula (XII):

wherein A, R₈ and W have the same meanings as defined above and then,

reacting the compound of the formula (XII) with a nitrogen-containing compound represented by the following formula (V):

 $\begin{array}{c}
\begin{pmatrix}
CH_2
\end{pmatrix}_n \\
HN \\
E_1
\end{array}
\begin{array}{c}
Y-(B)_m-D
\end{array}$ (V)

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

19. A process for the preparation of a compound represented by the following formula (Ij):

wherein A, B, D, E₁, E₂, Y, m and n have the same meanings as defined above, which comprises dehydrating a compound represented by the following formula (X):

 $\begin{array}{c}
OH \\
S \\
N \\
O_2
\end{array}$ (X)

wherein A and W have the same meanings as defined above, to obtain a compound represented by the following formula (XIII):

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wherein A and W have the same meanings as defined above and then reacting the compound of the formula (XIII) with a nitrogen-containing compound represented by the following formula (V):

$$(CH_2)_n$$

$$Y - (B)_m - D \qquad (V)$$

$$E_1 \quad E_2$$

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

20. A process for the preparation of a compound represented by the following formula (lk):

$$\begin{array}{c|c}
 & (CH_2)_n \\
 & & Y-(B)_m-D
\end{array}$$
(Ik)

wherein A, B, D, E₁, E₂, Y, m and n have the same meanings as defined above, which comprises reducing the compound represented by the following formula (XIII):

wherein A and W have the same meanings as defined above to obtain a compound represented by the following formula (XIV):

wherein A and W have the same meanings as defined above then reacting the compound of the formula (XIV) with a nitrogen-containing compound represented by the following formula (V):

$$HN \underbrace{\begin{array}{c} (CH_2)_n \\ Y-(B)_m-D \end{array}} (V)$$

wherein B, D, E₁, E₂, Y, m and n have the same meanings as defined above.

10 21. A process for the preparation of a compound represented by the following formula (XVII):

wherein R_1 has the same meaning as defined above, which comprises reacting a compound represented by the following formula (XV):

with a compound represented by the following formula (XVI):

wherein R₁ has the same meaning as defined above.

40 22. A process for the preparation of a compound represented by the following formula (XIX):

wherein R'₁ has the same meaning as defined above, which comprises reducing a compound represented by the following formula (XVIII):

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wherein R'₁ has the same meaning as defined above.

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23. An intermediate suitable for use in the production of a pharmaceutical product, said intermediate being represented by the following formula (XVII):

$$\begin{array}{c}
OR_1 \\
S\\
O_2
\end{array}$$
(XVII)

wherein R₁ has the same meaning as defined above.

24. An intermediate suitable for use in the production of a pharmaceutical product, said intermediate being represented by the following formula (XX):

wherein R₈ has the same meaning as defined above.

25. An intermediate suitable for use in the production of a pharmaceutical, said intermediate being represented by the following formula (XXI):

$$\begin{array}{c|c}
 & Z \\
 & N \\
 & N \\
 & O_2
\end{array}$$
(XXI)

wherein A, W and Z have the same meanings as defined above.

26. A serotonin-2 receptor antagonist, comprising as an effective ingredient a benzothiazine derivative or a salt thereof according to claim 1.

	27.	27. A therapeutic for circulatory diseases, comprising as an effective or a salt thereof according to claim 1.	e ingredient a benzothiazine derivativ
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP94/02194

A. CLASSIFICATION OF SUBJECT MATTER					
Int. C16 C07D279/02, C07D417/06, C07D513/10, A61K31/54					
According to International Patent Classification (IPC) or to both national classification and IPC					
B. FIELDS SEARCHED					
Minimum documentation searched (classification system followed b	y classification symbols)				
Int. Cl ⁶ C07D279/02, C07D417/		1/54			
Documentation searched other than minimum documentation to the	extent that such documents are included in t	he fields searched			
Electronic data base consulted during the international search (name CAS ONLINE	of data base and, where practicable, search	terms used)			
C. DOCUMENTS CONSIDERED TO BE RELEVANT					
Category* Citation of document, with indication, where	appropriate, of the relevant passages	Relevant to claim No.			
X FR, A, 2675801 (RHONE-POU Y October 30, 1992 (30. 10.		1-3, 7-11, 13, 25 1-3, 7-13, 25-27			
X JP, A, 311059 (Bayer AG.) Y December 15, 1989 (15. 12 & EP, A, 338331 & US, A,	. 89)	1, 7-10 1-3, 7-13, 25-27			
Y JP, A, 2-67274 (Rhône-Pou March 7, 1990 (07. 03. 90 & EP, A, 350430 & US, A,)	1-3, 7-12, 25-27			
Y WO, A, 93/16073 (THE WELL LIMITED), August 19, 1993 & EP, A, 625978		1-3, 7-12, 25-27			
Y JP, A, 255063 (Rhône-Poul November 13, 1991 (13. 11 & EP, A, 433149	· · · · · · · · · · · · · · · · · · ·	1-3, 7-12, 25-27			
X Further documents are listed in the continuation of Box C. See patent family annex.					
Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document bublished after the international filling date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be					
"E" cartier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance; the claimed inventor cannot be sup when the document is taken alone document of particular relevance; the claimed inventor cannot be					
"O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than					
the priority date claimed "&" document member of the same patent family					
Date of the actual completion of the international search March 7, 1995 (07. 03. 95) Date of mailing of the international search report March 28, 1995 (28. 03. 95)					
Name and mailing address of the ISA/	Authorized officer				
Japanese Patent Office					
Facsimile No.	Telephone No.				

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INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP94/02194

C (Continu	Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appropriate, of the rele	vant passages	Relevant to claim No.		
Y	Eur. J. Pharmacol., Vol. 204, No. 1, Roger Godbout et al., pp. 97-100	1991,	1-3, 7-10, 25-27		
Y	J. Med. Chem., Vol. 36, Apr. 1993, Jes Malleron et al., pp. 1194-1206	an-Luc	1-3, 7-10, 25-27		
Y	J. Med. Chem., Vol. 34, 1991, Jean-Luc Malleron et al., pp. 2477-2483	c	1-3, 7-10, 12, 25-27		
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